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# How to Reduce Charge Recombination in Organic Solar Cells: There are Still Lessons to Learn from P3HT:PCBM

Sebastian Wilken,\* Dorothea Scheunemann, Staffan Dahlström, Mathias Nyman, Jürgen Parisi, and Ronald Österbacka

Suppressing charge recombination is key for organic solar cells to become commercial reality. However, there is still no conclusive picture of how recombination losses are influenced by the complex nanoscale morphology. Here, new insight is provided by revisiting the P3HT:PCBM blend, which is still one of the best performers regarding reduced recombination. By changing small details in the annealing procedure, two model morphologies are prepared that vary in phase separation, molecular order, and phase purity, as revealed by electron tomography and optical spectroscopy. Both systems behave very similarly with respect to charge generation and transport, but differ significantly in bimolecular recombination. Only the system containing P3HT aggregates of high crystalline quality and purity is found to achieve exceptionally low recombination rates. The high-quality aggregates support charge delocalization, which assists the re-dissociation of interfacial charge-transfer states formed upon the encounter of free carriers. For devices with the optimized morphology, an exceptional long hole diffusion length is found, which allows them to work as Shockley-type solar cells even in thick junctions of 300 nm. In contrast, the encounter rate and the size of the phase-separated domains appear to be less important.

efficiencies approaching 20%.<sup>[1–3]</sup> However, despite the wide variety of materials that are now available, there are still only a few systems that maintain their full performance at junction thicknesses of 300 nm and more.<sup>[4–6]</sup> Compatibility with thick active layers, as well as a general tolerance to thickness variations, is considered an important prerequisite for the low-cost production of OPVs using printing techniques.<sup>[7,8]</sup> The main problem with increasing thickness is the slow-down of charge collection, which makes photogenerated carriers more vulnerable to recombination.<sup>[9–11]</sup> This is particularly true for NFA-based systems, which typically have low carrier mobilities of  $10^{-9}$  to  $10^{-8}$  m<sup>2</sup> V<sup>-1</sup> s<sup>-1</sup>. Hence, in order to compensate for the limitations in transport, it has become extremely important to find strategies how charge recombination can be suppressed.

Despite considerable efforts, there is still no agreement on which are the key factors determining the recombination strength.<sup>[12,13]</sup>

Conceptually, free charge recombination is a bimolecular process and can be described by the rate equation  $R = k_2 n^2$ , where  $k_2$  is the rate constant and  $n$  the carrier density. The parameter  $k_2$  is often compared with the homogeneous Langevin model,

$$k_L = \frac{q}{\epsilon \epsilon_0} (\mu_n + \mu_p) \quad (1)$$

where  $q$  is the elementary charge,  $\epsilon \epsilon_0$  the dielectric permittivity,  $\mu_n$  the electron mobility and  $\mu_p$  the hole mobility. Although there are a number of systems in which  $k_2$  is significantly reduced compared to Langevin recombination ( $k_2 = \zeta k_L$ , where  $\zeta < 1$ ), the reduction is usually not great enough to ensure thickness-insensitive device performance. It is generally accepted that the reduction factor  $\zeta$  is affected by the blend morphology, but the details remain controversial. For example, while some authors relate  $\zeta$  to the phase separation between the donor and acceptor,<sup>[14–18]</sup> others highlight the importance of phase purity and molecular order.<sup>[19–25]</sup> The difficulties of manipulating the morphology in a controlled way makes experimental clarification a complex task.

Among the benchmark systems for reduced recombination are blends based on the classical polymer

## 1. Introduction

The emergence of nonfullerene acceptors (NFAs) has pushed bulk-heterojunction organic photovoltaics (OPVs) to record

Dr. S. Wilken, S. Dahlström, Dr. M. Nyman, Prof. R. Österbacka  
Physics, Faculty of Science and Engineering  
Åbo Akademi University  
Porthansgatan 3, 20500 Turku, Finland  
E-mail: sebastian.wilken@abo.fi

Dr. S. Wilken, Dr. D. Scheunemann, Prof. J. Parisi  
Institute of Physics  
Carl von Ossietzky University of Oldenburg  
26111 Oldenburg, Germany

Dr. D. Scheunemann  
Department of Chemistry and Chemical Engineering  
Chalmers University of Technology  
412 96 Göteborg, Sweden

 The ORCID identification number(s) for the author(s) of this article can be found under <https://doi.org/10.1002/aelm.202001056>.

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poly(3-hexylthiophene) (P3HT). Both with fullerene and NFAs, reduction factors as low as  $10^{-4}$  have been reported.<sup>[14,26,27]</sup> In particular, the availability of NFAs that complement the absorption of P3HT and minimize voltage losses due to a fine-tuned energy level alignment has led to a remarkable renaissance of P3HT in OPV research.<sup>[27–34]</sup> One special feature of P3HT-based OPVs is that they develop their full performance only after the application of post-deposition treatments such as thermal and solvent annealing. The annealing induces phase separation and crystallization, thereby transforming the active layer into an “optimized” morphology in terms of charge generation and transport.<sup>[35–37]</sup> However, a closer look at the literature reveals that the connection between the morphological changes and the recombination is much less clear. In the Supporting Information (Table S1 and Figure S1) we collected 16 studies on annealed blends of P3HT and phenyl-C<sub>61</sub>-butyric acid methyl ester (PCBM), which is the most studied system to date. Even for nominally equally processed devices, reported reduction factors  $\zeta$  span over three orders of magnitude. Although the variation may be partly explained by different measurement methods, it points toward an underlying structure–property relationship that is yet to be revealed. Understanding this relation, or simply answering the question “what makes P3HT performing so well?”, will help in designing new materials for commercially relevant OPVs.

In this article, we revisit the P3HT:PCBM blend to determine the key morphological features to suppress charge recombination in OPVs. Specifically, we consider two model morphologies that behave similarly with respect to generation and transport, but show clear differences in recombination strength. By combining electron tomography, optical spectroscopy, and electrical measurements, we draw clear connections between morphology and the thickness-dependent competition between charge collection and recombination. We find that the presence of aggregates of high crystalline quality and purity is crucial for OPVs to function as Shockley-type devices without recombination losses even at high thickness. In contrast, the experiments show that the size of the phase-separated domains is of secondary importance. Our results suggest that delocalization along conjugated chain segments is key to allow charge-transfer pairs formed by the encounter of free electrons and holes to re-dissociate rather than to recombine into the ground state. This provides clear design rules for future OPV materials.

## 2. Results

### 2.1. Sample Systems and Device Performance

Throughout the following, we compare P3HT:PCBM films in 1:1 blend ratio that were spin-coated from either chlorobenzene (CB) or 1,2-dichlorobenzene (DCB) and subsequently thermally annealed. The main effect of the solvent is on the drying rate. While films from CB dried rapidly during spin coating, the ones from DCB were still wet after deposition and let dry slowly before the thermal anneal was applied.

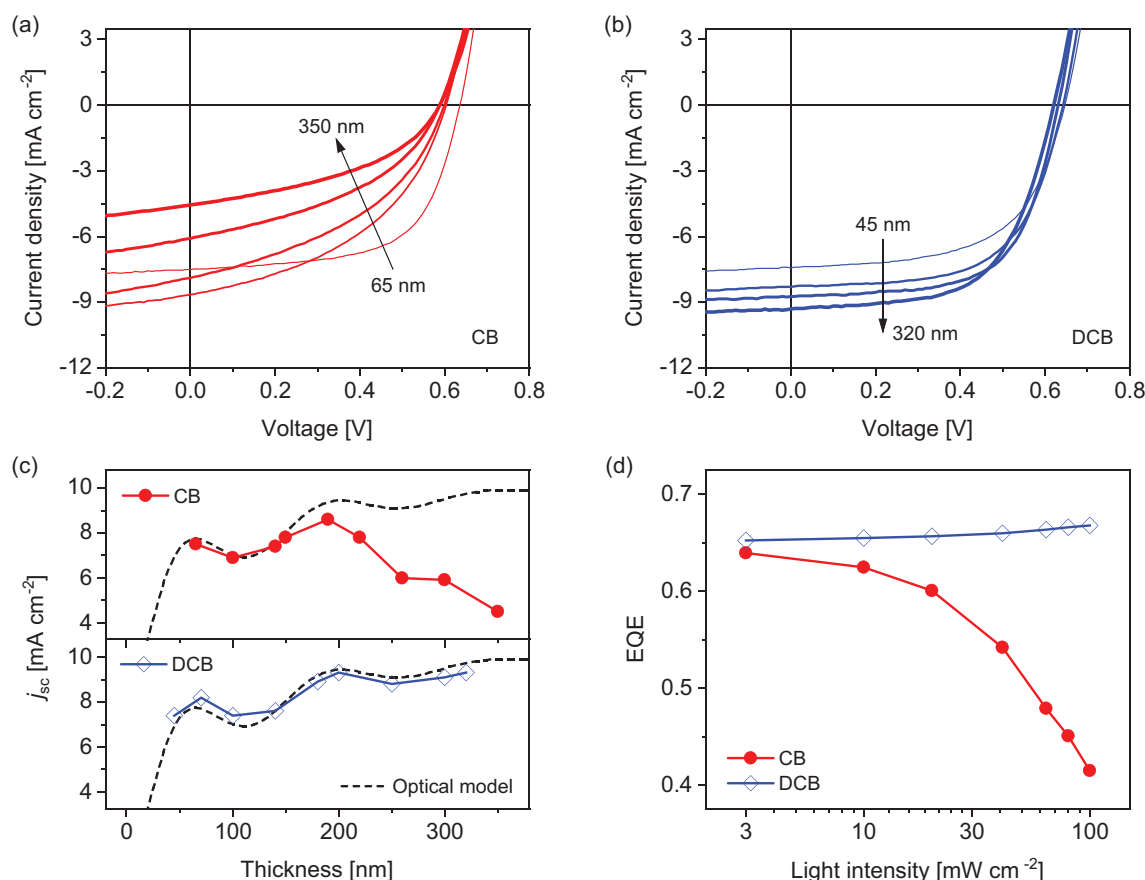
We implemented the differently prepared blends into inverted solar cells (substrate/cathode/active layer/anode) and varied the thickness  $L$  of the active layer. **Figure 1a,b** and Table S2, Supporting Information summarize the thickness-dependent current–voltage ( $j$ – $V$ ) characteristics under simulated solar illumination. While the performance of the CB and DCB systems is very similar in thin devices (efficiency around 3%), significant differences become apparent with increasing thickness. Clearly, only the slow grown DCB devices maintain (and even improve) their performance by increasing  $L$  from around 50 to over 300 nm. In contrast, for the CB devices, the photocurrent becomes increasingly voltage-dependent with increasing thickness, which leads to a clear drop in short-circuit current ( $j_{sc}$ ) and fill factor (FF). As a result, the thickest CB device ( $L = 350$  nm) achieves an efficiency of only 1.1%.

Figure 1c shows that the photocurrent of the DCB devices is well described by an optical transfer-matrix model assuming an internal quantum efficiency (IQE) independent on thickness. We determined the IQE experimentally using the external quantum efficiency (EQE) measured with lock-in technique (see Figure S3, Supporting Information). In the case of the CB devices, this method fails to explain the drop of  $j_{sc}$  for  $L > 150$  nm. The reason is that the EQE decreases strongly with increasing light intensity, which can be seen by adding bias illumination to the low-intensity probe in the EQE measurement (Figure 1d). Hence, the poor performance is not an inherent property of the thick CB devices, but develops gradually with increasing carrier density. This aspect is further elaborated in Figures S4 and S5, Supporting Information, where light-intensity dependent  $j$ – $V$  curves and EQE spectra are shown. Such a behavior is usually attributed to bimolecular recombination, often accompanied by space-charge effects.<sup>[38]</sup> Importantly, at low intensity, all CB and DCB devices display a rather high IQE of about 70%. This illustrates that free charge carriers are efficiently generated in both systems, while the differences lie in how the collection of those carriers competes with recombination.<sup>[9,11]</sup>

Having shown that seemingly small details in the blend preparation have drastic effect on the thickness-dependent device performance, we now want to establish relationships with the morphology. For this purpose, we will first present a detailed characterization of the relevant structural features, that is, phase separation, aggregation, and phase purity. We then use mobility measurements, as well as transient photocurrent and photovoltage studies to relate these properties to the kinetics of charge collection and recombination.

### 2.2. Phase Separation

To investigate the phase separation in real space, we used high-resolution transmission electron microscopy (TEM) and tomography (**Figure 2**). Blend films of  $\approx 200$  nm thickness were selected for this analysis because they are representative of the thickness series and still exhibit high enough electron transparency.<sup>[39,40]</sup> **Figure 2a,c** show regular bright-field TEM images, which reveal significant differences at various length



**Figure 1.** Thickness-dependent performance of CB and DCB devices. a,b)  $j$ - $V$  curves under simulated sunlight for different active-layer thicknesses. See Table S2, Supporting Information for the full data set. c) Measured  $j_{sc}$  (symbols) together with the result of transfer-matrix calculations (dashed lines) assuming a constant IQE of 70%. d) White-light bias dependent EQE for 300-nm thick devices.

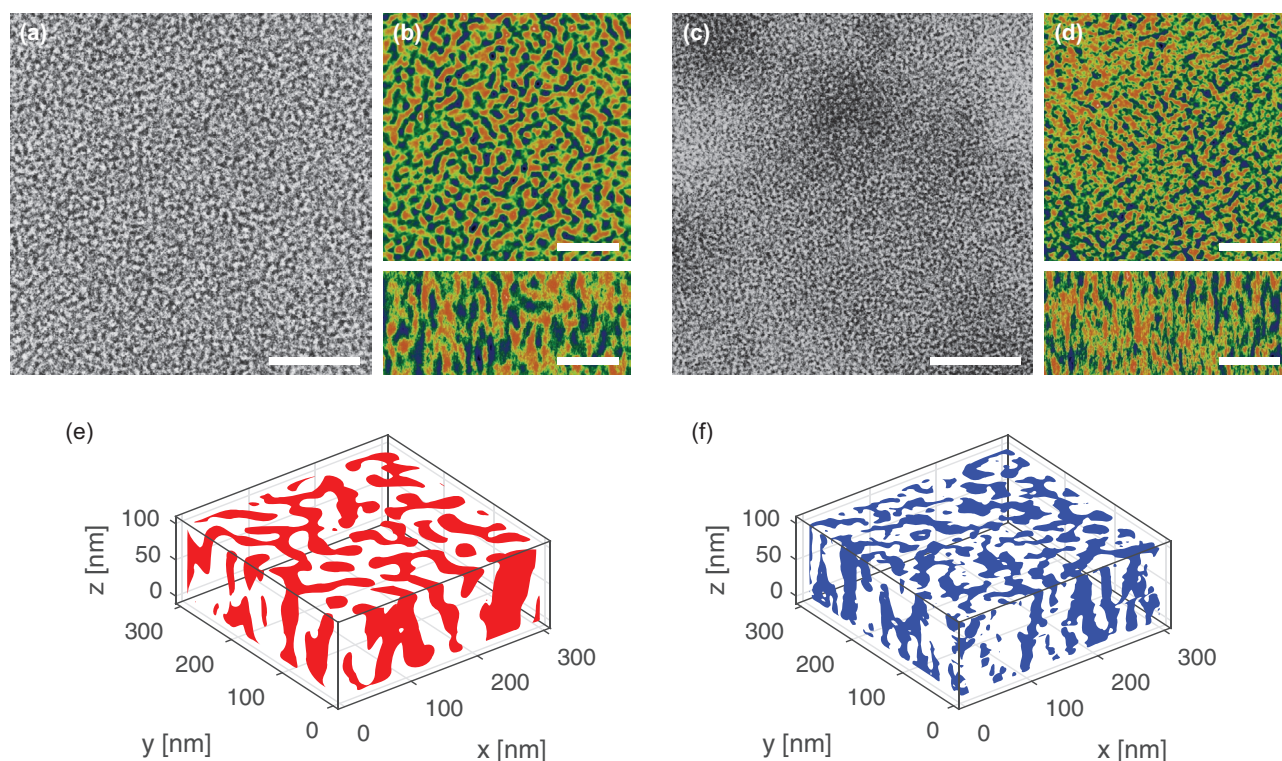
scales. While the CB sample exhibits no distinct structural features on the micrometer scale, alternating regions of bright and dark contrast can be seen for the DCB sample. Atomic force microscopy (Figure S6, Supporting Information) confirms that these alternations are due to the surface topography (i.e., height variations) rather than the blend composition. Independent of the substrate used, much smoother films with a root-mean-square roughness of  $\approx 1$  nm were obtained by rapid drying (CB), as compared to  $\approx 10$  nm in the case of slow drying (DCB), in agreement with previous works.<sup>[37,41]</sup>

Instead of that, image contrast on the nanometer scale contains information about the phase separation. Because of the lower density of P3HT ( $\approx 1.1 \text{ g cm}^{-3}$ ) compared to PCBM ( $1.3 \text{ g cm}^{-3}$ ), bright areas can be assigned to polymer domains and dark areas to fullerene domains.<sup>[42–45]</sup> The regular TEM images already suggest a coarser phase separation (i.e., larger domains) for the CB cast blend films. However, as these are 2D projections of a 3D network, they are insensitive to possible vertical gradients.<sup>[44,46]</sup> To overcome this limitation, we used electron tomography, which yields a volumetric reconstruction of the blend film. Figure 2b,d show exemplary slices through the tomograms parallel ( $xy$  direction) and perpendicular ( $xz$  direction) to the film plane. The tomographic data confirms the trend from the regular TEM imaging. While the CB sample exhibits well-separated and homogeneously distrib-

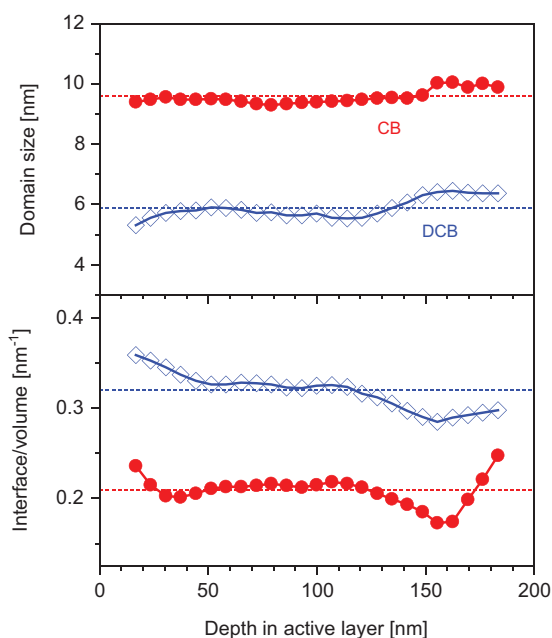
uted domains in the order of 10 nm, the DCB sample shows a much finer phase separation with a higher degree of interpenetration between the P3HT and PCBM domains. From the  $xz$  slices, a columnar structure with transport paths toward the bottom and top of the film can be seen in both cases. Figure 2e,f give an impression of the 3D phase-separated morphology. For this representation, the slices were binarized by attributing each pixel either to the P3HT or PCBM phase (see also Movie S1, Supporting Information).

For a quantitative statistical analysis of the structural features, we computed the 2D autocorrelation using the fast Fourier transform (FFT) algorithm as detailed in the Supporting Information. The upper panel of Figure 3 shows the domain size  $d$ , estimated from the width of the autocorrelation function, versus the vertical position. In case of the rapidly dried sample (CB), the domain size was found to be nearly constant throughout the film, with an average value of  $d = 9.6 \pm 0.2$  nm. In case of the slowly grown sample (DCB), the domain size was slightly increasing (from 5.3 to 6.5 nm) toward the upper boundary, with an average of  $d = 5.9 \pm 0.4$  nm. From the binarized slices, we were further able to compute Minkowski functionals such as perimeter and area.<sup>[47]</sup> We used these measures to estimate the interfacial-area-to-volume ratio (Figure 3, lower panel). As one would expect, the finer phase separation of the DCB sample is accompanied by a larger amount of interfacial





**Figure 2.** Morphology of 200 nm thick P3HT:PCBM films processed from CB (panels a,b and e) and DCB (panels c,d and f), respectively. a,c) Regular bright-field TEM images of free-standing films. Scale bar: 300 nm. b,d) Exemplary slices through electron tomographic reconstructions parallel (top) and perpendicular (bottom) to the film plane. The color coding represents the brightness value of a certain pixel, which decreases from red over green to blue. Scale bar: 100 nm. e,f) Representative volume elements, reassembled from the xy slices after binarization. See also Movie S1, Supporting Information.

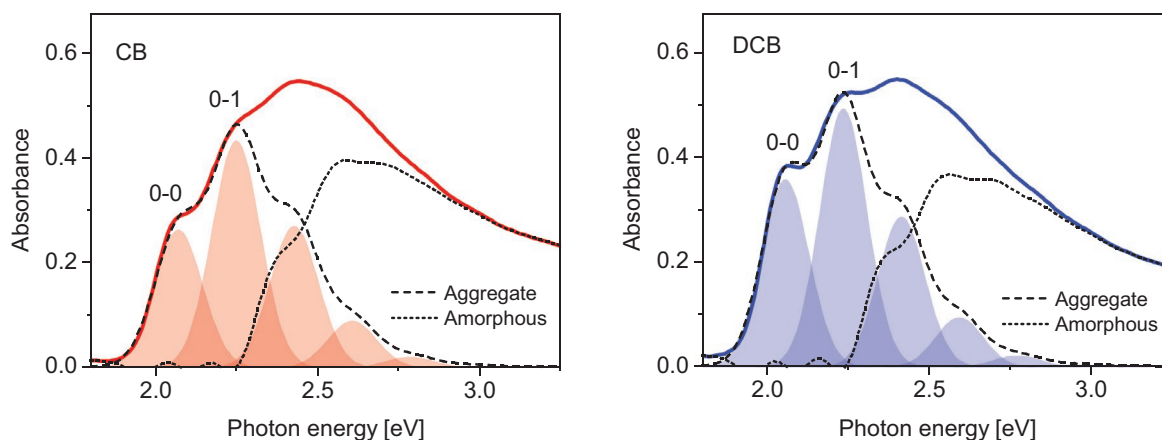


**Figure 3.** Morphological features derived from the electron tomograms. Upper panel: Domain size  $d$  estimated from the width of the self-correlation peak of the radially averaged 2D autocorrelation function (see the Supporting Information). Lower panel: Interfacial-area-to-volume ratio of the P3HT phase calculated using Minkowski functionals. Dashed lines indicate the average values.

area. We note that due to the 1:1 blend ratio by weight, the P3HT and PCBM phases are not equal in volume. From the number of bright to dark pixels, we estimate the P3HT volume fraction to 0.57 (CB) and 0.54 (DCB), respectively, which fits well to the density ratio mentioned above. Furthermore, the slightly lower P3HT volume for the DCB sample hints to a higher degree of aggregation, as will be discussed in the next section.

### 2.3. Aggregation and Phase Purity

The casting solvent does not only influence the size of the phase-separated domains, but also the molecular order within them.<sup>[46,48,49]</sup> To complement our TEM studies with information about the internal domain structure, we used optical absorption spectroscopy (Figure 4). Generally, blend films cast from DCB showed more pronounced 0–0 and 0–1 vibronic features, suggesting a higher ordering in the polymer phase.<sup>[50,51]</sup> It is known that the P3HT phase in P3HT:PCBM blend films consists of a mixture of amorphous and aggregated material, the latter formed by lamellar crystallites of 2D conjugated sheets.<sup>[51–53]</sup> For a spectral decomposition of the absorption bands, we fitted the P3HT absorbance component to the model of weakly interacting H-aggregates by Spano and coworkers.<sup>[54–56]</sup> The model treats the absorption in the ordered regions by a series of Gaussian peaks, determined by three parameters: the energy



**Figure 4.** Decomposition of the absorption spectra of CB (left) and DCB (right) cast blends after thermal annealing. Solid lines are the P3HT component of the P3HT:PCBM absorbance. Dashed lines are fits to the Spano model in the aggregate region (1.95 to 2.25 eV), and shaded areas are the single Gaussian contributions (see Experimental Section for fitting details). Dotted lines represent the residual absorption attributed to amorphous P3HT.

$E_{0-0}$  of the 0-0 transition, the Gaussian bandwidth  $\sigma$ , and the intramolecular free exciton bandwidth  $W$  (see Experimental Section for details).

**Table 1** and the dashed lines in Figure 4 show the results of the Spano analysis for the spectral region dominated by aggregate absorption. The main difference between the CB and DCB cast blends lies in the exciton bandwidth  $W$ , which is known to scale inversely with the intramolecular order in the P3HT aggregates.<sup>[57,58]</sup> In particular,  $W$  has been correlated with the conjugation length, that is, the number of interacting thiophene repeat units. We can estimate the conjugation length using previous approaches,<sup>[41,58,59]</sup> which gives about 27 repeat units for the CB blends and 40 repeat units for the DCB blends. Since both systems were subjected to the same thermal annealing, the apparent differences can only be explained by the solvent evaporation rate. We hypothesize that the fast evaporation of CB gives the P3HT not enough time to organize into larger aggregates, and that thermal annealing cannot fully convert the disordered structure into a highly ordered one with an extended conjugation. This is different to the DCB blends, where the slow drying already leads to the formation of aggregates of high crystalline quality.<sup>[41]</sup> In contrast,  $E_{0-0}$  and  $\sigma$  are largely unaffected by the solvent. This means that the aggregates in the CB and DCB blends have very similar energetic properties, despite their significantly different expansion in the direction of conjugation. Interestingly, very similar trends with the solvent have recently been reported for a P3HT:NFA blend.<sup>[60]</sup> Therefore, we assume that our results are generally valid for P3HT and similar semicrystalline systems.

**Table 1.** Results of the Spano analysis for CB and DCB cast blend films.

Solvent	$E_{0-0}$ [eV]	$\sigma$ [meV]	$W$ [meV]
CB	2.048	75	132
DCB	2.040	73	87

From the difference between measurement and Spano fit we can also calculate the amorphous absorption component (Figure 4, dotted lines). Using the ratio between the two absorption fractions, and the extinction coefficients of P3HT in the aggregated and amorphous state,<sup>[56]</sup> we estimate the aggregate percentage in the P3HT phase to 45% (CB) and 51% (DCB), respectively. Even if we cannot resolve the aggregation in our tomography data, there is a direct link to these numbers: Because of the closer packing, aggregated P3HT has a higher density than amorphous P3HT.<sup>[61]</sup> Hence, at a given weight, a slightly smaller P3HT volume is expected in the DCB blends with the higher aggregate percentage. This is exactly the trend we derived from the tomography analysis.

When blended with an acceptor, the P3HT aggregation is directly correlated with the phase purity. In particular, PCBM is known to be intermiscible with P3HT in amorphous state, but not with P3HT in aggregated state.<sup>[62]</sup> If we assume that all amorphous P3HT is molecularly mixed with PCBM, we get a refined picture of the morphology consisting of roughly one third pure P3HT, one third pure PCBM and one third mixed phase. For a more accurate estimate, we can combine the P3HT volume from the tomography measurements with the amorphous P3HT fraction from the Spano analysis. This way we arrive at about 31% mixed phase in the CB blends and 26% in the DCB blends, which lies in the range that has been reported for annealed P3HT:PCBM blends using high-resolution spectroscopic imaging.<sup>[63,64]</sup> Hence, we can conclude that in addition to the higher intramolecular order in the P3HT aggregates, the DCB blends also exhibit a higher overall phase purity.

## 2.4. Charge Transport

Having shown that the CB and DCB blends show clear differences in their morphological features, we now turn to the electrical properties. To estimate the carrier mobilities, we used space-charge limited current (SCLC) experiments. Figure S8,

**Table 2.** Charge transport parameters derived from SCLC measurements.

	Electrons		Holes	
	CB	DCB	CB	DCB
Mobility, $\mu$ [ $\text{m}^2 \text{V}^{-1} \text{s}^{-1}$ ]	$1.5 \times 10^{-7}$	$8.9 \times 10^{-8}$	$1.6 \times 10^{-8}$	$1.3 \times 10^{-8}$
Gaussian disorder, $\sigma$ [meV]	107	112	79	63
Attempt-to-hop frequency, $\nu_0$ [ $\text{s}^{-1}$ ]	$3 \times 10^{12}$	$2 \times 10^{12}$	$9 \times 10^9$	$7 \times 10^9$

Supporting Information shows  $j$ - $V$  curves of electron-only and hole-only devices whose active layers were prepared the same way as for the solar cells. As the simplest and most robust model, we fitted the data in the SCLC regime to the Mott-Gurney law, and Table 2 lists the resulting electron and hole mobilities. As the most important result, the magnitude of both  $\mu_n$  and  $\mu_p$  appears to be fairly independent of the solvent used.

We further analyzed the data in terms of the Gaussian disorder model (GDM), which explicitly takes into account the hopping nature of transport.<sup>[65,66]</sup> Also the derived hopping parameters given in Table 2, that is, the Gaussian disorder  $\sigma$  and the attempt-to-hop frequency  $\nu_0$ , point to very similar transport properties between the CB and DCB blends. Notably, while there is virtually no difference in the Gaussian disorder for electrons, the disorder for hole transport is  $\approx 15$  meV smaller in case of the DCB samples. This shows once again that it is mainly properties of the P3HT phase that are influenced by the solvent. Considering that the apparent transport characteristics are an average over aggregated and amorphous regions, the result is also in line with our hypothesis that the DCB blends consist of slightly less disordered material. Macroscopically, however, the beneficial effect of the narrower density of states (lower  $\sigma$ ) holes are experiencing in the DCB blends will be canceled out by the slower hopping rate (lower  $\nu_0$ ), so that the results from the GDM model are overall consistent with the Mott-Gurney analysis. This is also in agreement with the general assumption that the macroscopic transport properties are largely dominated by the transport through amorphous regions (which we assume are not significantly different between the CB and DCB blends) rather than within the aggregates.<sup>[41,59,67]</sup>

It is important to note that SCLC diodes probe the transport characteristics of charges injected from the contacts. To check whether the results are also relevant for photogenerated charges, we performed resistance-dependent photovoltage (RPV) measurements on operational solar cells devices. The RPV method is a transient technique with ns to ms resolution, that is, the time scale relevant for charge collection and recombination.<sup>[68]</sup> From the RPV transients shown in Figure S9, Supporting Information, very similar electron mobilities can be derived as from the SCLC measurements, and about two times lower hole mobilities. As RPV is carried out under much lower carrier densities, the latter might point to a slight carrier density dependence of  $\mu_p$ , but is within the typical range when comparing different mobility measurements methods.<sup>[69]</sup> However, the important fact is that there is still no significant difference between the CB and DCB blends. This shows that also the (macroscopic) transport of photogenerated charges is only slightly influenced by the differences in morphology. In particular, both the CB

and DCB blends show the typical mobility imbalance of about one order of magnitude.<sup>[70,71]</sup> Consequently, the thick devices are supposed to be affected by space-charge effects, as will be discussed below.

## 2.5. Charge Collection Versus Recombination

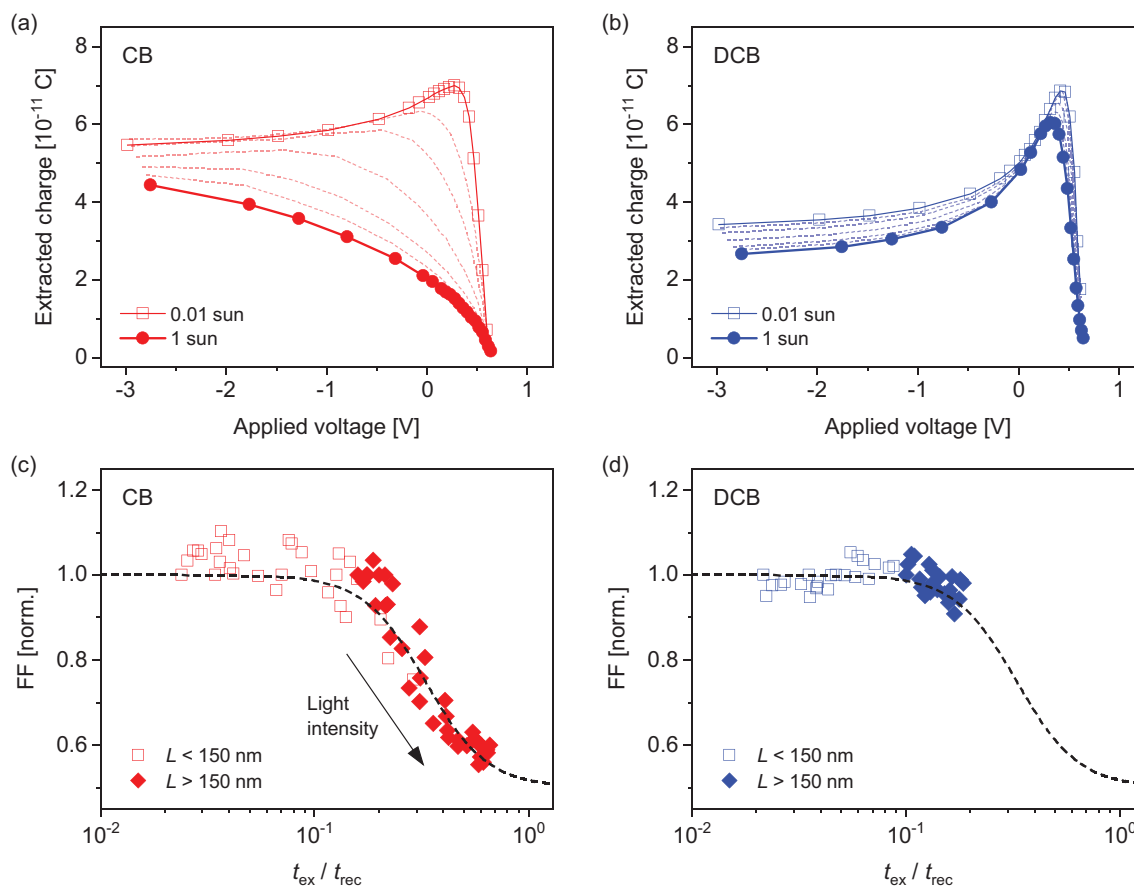
We now focus in more detail on the thickness-dependent competition between charge collection and recombination. To study the dynamics of collection, we measured the transient photocurrent (TPC) due to a small optical perturbation while the device is held at a constant bias voltage and background illumination. We used a relatively long pulse length of 100  $\mu\text{s}$  to guarantee that a steady state is reached. Because of the finite carrier mobility, a transient current  $j(t)$  is observed after switching off the light pulse at time  $t = 0$ , which reflects the carrier sweep-out.<sup>[72,73]</sup> Figure 5a,b illustrates the TPC behavior of 300-nm devices by plotting the voltage dependence of the extracted charge,

$$\Delta Q = \int_0^{t_f} j(t) dt \quad (2)$$

where  $t_f$  is a time at which charge collection is completed. Data for the whole thickness series can be found in Figures S10 and S11, Supporting Information.

At low background illumination, all devices exhibit a similar behavior, which can be understood as follows. Reducing the internal voltage,  $V_{\text{int}} = V_{\text{oc}} - V$ , by going from reverse to forward bias slows down the current decay and  $\Delta Q$  becomes larger. However, as  $V \rightarrow V_{\text{oc}}$ , the internal field is close to zero and the extracted charge is reduced due to recombination. Altogether, this leads to characteristic maximum of  $\Delta Q$ , indicating the point at which recombination starts to compete with collection. Such a voltage dependence proves that the TPC signals represent the free charge carrier dynamics and are not limited by RC effects.<sup>[73]</sup> The situation changes with increasing background illumination. In the thicker CB devices ( $L > 150$  nm), the extracted charge is drastically reduced and the maximum of  $\Delta Q$  shifts gradually toward higher internal fields. Hence, recombination clearly competes with collection over a large range of voltages. This is in contrast to the DCB devices, where  $\Delta Q$  is nearly invariant to thickness and light intensity. Also in the thick DCB devices, recombination is only significant close to  $V_{\text{oc}}$ , where photogenerated carriers mostly recombine with carriers injected from the contacts.<sup>[74]</sup>

Figure 5c,d illustrates the relevance of the TPC dynamics for the solar cell performance. Shown is the device FF versus the ratio between the TPC decay time  $t_{\text{ex}}$  in the extraction regime (reverse bias) and  $t_{\text{rec}}$  in the recombination regime (close to  $V_{\text{oc}}$ ) for a range of thicknesses and light intensities. The ratio  $t_{\text{ex}}/t_{\text{rec}}$  serves as a figure of merit for the competition between collection and recombination.<sup>[9,11]</sup> Notably, all data points collapse into a universal curve. For the CB devices, the fill factor drops when  $t_{\text{ex}}$  and  $t_{\text{rec}}$  are in the same order of magnitude, which is the case for  $L > 150$  nm at high light intensities. In contrast, for the DCB devices, the absence of data



**Figure 5.** Dynamics of charge collection derived from TPC measurements. a,b) Voltage dependence of the extracted charge  $\Delta Q$  for 300-nm thick devices at different background light intensities. c,d) Device FF versus the ratio  $t_{\text{ex}}/t_{\text{rec}}$  between the carrier response time in the extraction regime ( $V = -1\text{V}$ ) and the recombination regime ( $V \rightarrow V_{\text{oc}}$ ). Data points correspond to eight samples of different thickness  $L$  and background light intensities ranging from 0.01 to 1 sun. The fill factor is normalized to its low-intensity value to exclude other influences such as the quality of the contacts. Dashed lines are a guide to the eye.

points in this region indicates that collection is always faster than recombination. Given the very similar mobilities, it is not likely that carriers are collected at a higher rate. Instead, the striking differences between the two systems must be related to the charge recombination.

In order to characterize the recombination mechanism, we determined the reaction order  $\delta$  and the ideality factor  $n_{\text{id}}$  using transient and steady-state photovoltage measurements as described in the Supporting Information. For the CB devices, we find  $\delta$  close to 2 and  $n_{\text{id}}$  close to 1, which indicates that bimolecular recombination between free electrons and holes is the dominant loss mechanism.<sup>[75,76]</sup> Hence, we can employ the rate equation  $R = k_2 n^2$  and estimate the rate constant to  $k_2 \approx 2 \times 10^{-18} \text{ m}^3 \text{ s}^{-1}$  (Figure S13, Supporting Information). This value confirms recent charge extraction measurements on the same system,<sup>[38]</sup> which also show only a weak dependence on the carrier density (Figure 6a).

For the DCB devices, the apparent recombination behavior is more complex. We find  $\delta$  significantly exceeding 2 and  $n_{\text{id}}$  ranging between 1 and 2, which suggests that recombination involves carriers trapped in exponential tails of the density of states.<sup>[77]</sup> Given that the DCB blends actually consist of less disordered material than the CB blends, it does not seem

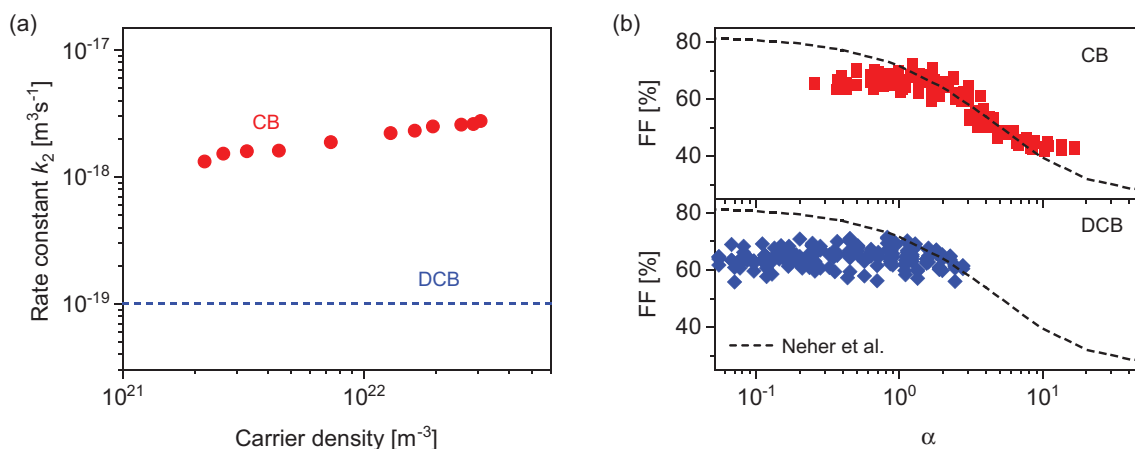
likely that they have more or deeper tail states. Instead, we assume the free carrier recombination (as given by  $k_2$ ) to be much stronger reduced, so that the trap-assisted regime becomes more apparent. Under these conditions, the transient techniques used herein to determine  $k_2$  do not lead to meaningful results.<sup>[78,79]</sup> However, to explain the differences in device performance, we estimate that  $k_2$  must be at least one order of magnitude smaller than in the CB blends. This is supported by a recent study in which the newly developed impedance-photocurrent device analysis technique showed a rate constant  $k_2$  of about  $10^{-19} \text{ m}^3 \text{ s}^{-1}$  for comparably processed DCB devices.<sup>[80]</sup>

To test whether a contrast in  $k_2$  alone can explain the striking differences between CB and DCB devices, we applied the modified Shockley equation by Neher et al.,<sup>[11]</sup>

$$j = qGL \left\{ \exp \left[ \frac{q}{(1+\alpha)k_{\text{B}}T} (V - V_{\text{oc}}) \right] - 1 \right\} \quad (3)$$

where  $G$  is the generation rate calculated with our transfer-matrix model and  $\alpha$  is a factor that relates charge generation, transport, and recombination to each other,





**Figure 6.** Characterization of the recombination kinetics. a) Rate constants  $k_2$  for the CB and DCB blends. The data points for the CB system were taken from a recent literature study<sup>[38]</sup> and confirmed by transient photovoltage measurements (Figure S13, Supporting Information). For the DCB system,  $k_2$  was estimated as detailed in the text. b) Device FF versus the parameter  $\alpha$  as given in Equation (4) for CB and DCB solar cells of various thickness and measured at different light intensities. Dashed lines are the expectations according to the modified Shockley model by Neher et al.<sup>[11]</sup> for recombination rate constants of  $k_2 = 2 \times 10^{-18} \text{ m}^3 \text{ s}^{-1}$  (CB) and  $k_2 = 1 \times 10^{-19} \text{ m}^3 \text{ s}^{-1}$  (DCB), respectively. The degradation of the FF in the CB devices can be explained by the larger  $k_2$  alone.

$$\alpha = \frac{q(k_2 G)^{1/2} L^2}{2\mu_p k_B T} \quad (4)$$

In the denominator we used only the mobility of the slower carrier (here: holes), which dominates the photocurrent if transport is significantly imbalanced.<sup>[11,38]</sup> Figure 6b shows the analytical relationship between the FF and  $\alpha$  (dashed lines) together with  $\approx 200$  data points each for the CB and DCB system, corresponding to different samples of variable thickness  $L$  measured at varying light intensity (and thus  $G$ ). There is a reasonable agreement between the Neher model and the experiments, which confirms the validity of the  $k_2$  values. We note that the deviations at low and high  $\alpha$  values are due to electrical imperfections (finite shunt resistance, surface recombination) in the thin devices and space-charge effects in the thick devices, respectively. An important finding from Figure 6b is that under nearly all conditions tested, the DCB devices operate as Shockley-type solar cells ( $\alpha < 1$ ). This is an outstanding result for OPVs, especially with thick active layers.<sup>[4]</sup> In contrast, most CB devices are in the transport-limited regime ( $\alpha > 1$ ).

Figure S14, Supporting Information shows that the Neher model can also correctly describe full  $j$ - $V$  curves at different light intensities. The notable exception is the thick CB devices at high intensities. The reason is that under these conditions the photocurrent becomes space-charge limited, which is not considered in Equation (3). It is important to note that due to the similar mobility imbalance, space charge will also build up in the thick DCB devices, as can be seen from modeled energy band diagrams (Figure S15, Supporting Information). We have recently shown that collection is limited to the width  $w$  of the space-charge region plus the diffusion length  $L_D$  of the slower charge carrier.<sup>[38]</sup> For the thick devices, we find  $w \approx 160 \text{ nm}$  under 1-sun illumination. This gives an alternative way to determine an upper limit for  $k_2$  in the DCB blends: The fact that charges are collected

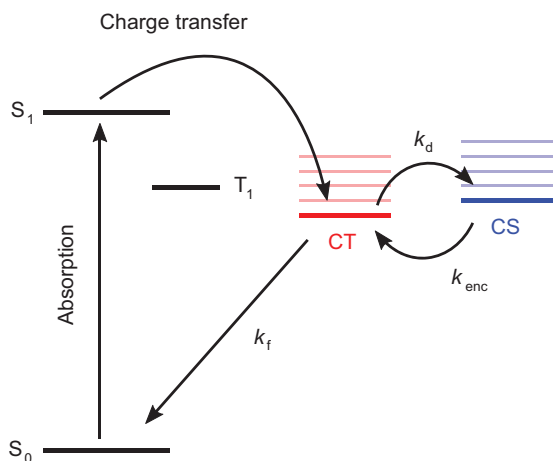
from a 300-nm device without noticeable loss means that the hole diffusion length must be at least around 140 nm. This is an exceptionally long diffusion length for OPVs, outperforming for instance the highly efficient PM6:Y6 system.<sup>[81]</sup> Using the relationship

$$L_D = \sqrt{\frac{\mu_p \tau k_B T}{q}} \quad (5)$$

where  $\tau = (k_2 G)^{1/2}$ , we can then estimate that  $k_2$  must be about  $1 \times 10^{-19} \text{ m}^3 \text{ s}^{-1}$  or less. This is in excellent agreement with our above assumption and confirms that the key difference between the CB and DCB devices lies in the free charge recombination, which is about 20 times more reduced in the DCB blends.

### 3. Discussion

We now want to discuss our findings in the light of recent recombination models and derive design rules for commercially relevant OPV materials. **Figure 7** summarizes the current understanding of charge generation and recombination in organic solar cells.<sup>[12,19,82–84]</sup> Both processes involve bound excitons (either in spin singlet or triplet state), less bound charge transfer (CT) pairs, and free carriers. Nongeminate charge recombination, on which we will focus in the following, is a two-step process. The first step is the encounter of a free electron and a free hole originating from different photoexcitations (rate constant  $k_{\text{enc}}$ ); the resulting encounter complex has been identified as CT state with similar properties to the one involved in charge generation.<sup>[83,85]</sup> The second step is the decay of the CT state into the ground state (rate constant  $k_f$ ), that is, the actual recombination event. However, instead of decaying into the ground state, the CT state also has the possibility to dissociate again (rate constant  $k_d$ ). Following this rationale, the



**Figure 7.** Illustration of the relevant energy levels and transitions for charge generation and recombination in organic bulk-heterojunction solar cells. The nongeminate recombination of electrons and holes from the charge separated (CS) state to the electronic ground state involves the formation of a charge transfer (CT) pair as intermediate.

experimentally observable bimolecular rate constant  $k_2$  can be written as

$$k_2 = \zeta_{CT} k_{enc} \quad (6)$$

where  $\zeta_{CT}$  is a reduction factor related to the CT kinetics.

When the decay of the CT state is much faster than its dissociation ( $k_f \gg k_d$ ), the recombination is encounter-limited and  $\zeta_{CT} \rightarrow 1$ . Hence, in this case, the recombination rate is solely given by  $k_{enc}$ . While in a homogeneous medium,  $k_{enc}$  should be given by the Langevin prefactor  $k_L$ , it may become reduced in a blend with phase separation. This is because electrons and holes are confined to different material phases and can only meet at the heterointerface.<sup>[14–16,86]</sup> Heiber et al.<sup>[17]</sup> provided a semi-analytical model for the encounter rate in the presence of phase separation,

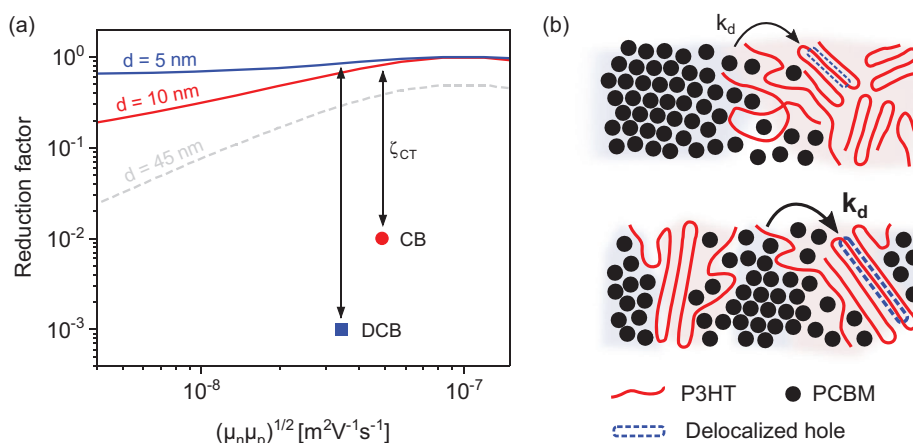
$$k_{enc} = \frac{q}{\epsilon \epsilon_0} 2 f_1 \left( \frac{\mu_n^g + \mu_p^g}{2} \right)^{1/g} \quad (7)$$

where  $f_1$  and  $g$  are domain-size dependent factors derived from Monte Carlo simulations on artificial blend morphologies. **Figure 8a** illustrates the possible reduction through Equation (7) for a range of mobilities and domains sizes. Comparing the numbers with the measured reduction factors yields two conclusions: First, phase separation cannot explain the differences between the CB and DCB blends. With the given mobilities and domain sizes, the Heiber model would predict recombination to be (slightly) weaker in the CB blends, which is the opposite trend to our experimental result. Second, the calculated encounter rates exceed the measured values of  $k_2$  by orders of magnitude, which proves that both the CB and DCB blends are not in the encounter-limited regime. Even in extreme cases, only relatively mild reductions ( $>10^{-2}$ ) are predicted, which shows that tuning the domain size is not a promising strategy to strongly suppress charge recombination.

By inserting the calculated encounter rates into Equation (6) we can deduce CT reduction factors of  $\zeta_{CT} = 10^{-2}$  and  $10^{-3}$  for the CB and DCB blends, respectively. In other words, only 1 in 100 or 1000 encounter events will lead to an actual loss, which implies that the probability for the CT state to separate again must be much higher than the probability to relax to the ground state ( $k_d \gg k_f$ ). As shown by Burke et al.,<sup>[19]</sup> this leads to an equilibrium between CT states and free carriers, reducing the recombination rate by

$$\zeta_{CT} = \frac{k_f}{k_f + k_d} \quad (8)$$

Several studies have indicated that the balance between  $k_d$  and  $k_f$  is moved toward separation by the presence of aggregates.<sup>[19,21,23,87,88]</sup> Energetically speaking, aggregation shifts the molecular orbitals such that the electronic gap is reduced compared to the amorphous state.<sup>[89]</sup> In a three-phase morphology,



**Figure 8.** Effect of morphology on charge recombination. a) Measured reductions  $k_2/k_L$  for the CB and DCB blends (data points) in comparison with the model by Heiber et al.<sup>[17]</sup> assuming encounter-limited recombination with a reduction factor  $k_{enc}/k_L$  (lines). Different traces belong to different domain sizes. b) Schematic illustration of the morphologies studied herein (top: CB, bottom: DCB). Our data suggests that the more extended delocalization of holes in the DCB blends assists CT states to dissociate.

consisting of molecularly mixed, amorphous regions and pure aggregates, this creates an energy cascade pushing carriers from mixed to pure regions. The additional driving force has been shown to not only improve the split-up of CT pairs during photogeneration, but also to reduce the loss due to charge recombination.<sup>[19,23]</sup>

Given the well documented energy level offset between P3HT aggregates and amorphous P3HT mixed with PCBM,<sup>[88]</sup> the significant P3HT aggregation in the CB and DCB blends reasonably explains why both systems are non-Langevin systems. However, the relatively small difference in the P3HT aggregate percentage of 45 to 51% alone cannot explain why the reduction is one order of magnitude stronger in the DCB blends. We also find no evidence that the properties of the PCBM are substantially altered by the solvent. Hence, it is reasonable to suggest that the crystalline quality in the P3HT aggregates, expressed by the exciton bandwidth  $W$ , is the decisive factor. Although the parameter  $W$  mainly refers to excitons, a direct connection to the charge domain could be shown.<sup>[56]</sup> Here, the significantly smaller value of  $W$  for the DCB blends translates into a larger number of interacting repeating units in the polymer. As a result, the holes are supposed to be more delocalized in the DCB blends. In other words, they have a higher local mobility (as opposed to the macroscopic mobility addressed in the transport measurements), which leads to a higher probability for carriers to overcome the energetic barrier between the interfacial CT state and the charge-separated state. This in turn increases  $k_d$  and thereby the denominator in Equation (8). For a schematic illustration of this scenario, see Figure 8b.

It should be noted that the CT dissociation rate will also affect the way how free charges are generated following photoexcitation. Hence, one might argue that the contrast in  $\zeta_{CT}$  is at odds with the similar generation efficiencies we found for the CB and DCB blends. However, as pointed out by Shoaee et al.,<sup>[82]</sup> equilibrium between CT states and free charges implies a reverse relationship of the form  $\zeta_{CT} = 1 - \eta_{CT, diss}$ , where  $\eta_{CT, diss}$  is the yield of CT dissociation during charge generation. Varying  $\zeta_{CT}$  from  $10^{-3}$  to  $10^{-2}$ , which has drastic consequences for the recombination behavior as shown in this work, corresponds only to a change in  $\eta_{CT, diss}$  from 0.999 to 0.99. Such small differences in generation efficiency are not distinguishable with the methods used herein. In other words, recombination is much more sensitive to the CT dissociation rate than generation is.

Another aspect to consider is the role of spin in the recombination. In general, the encounter of two independent charges should form CT states of singlet ( $^1CT$ ) and triplet ( $^3CT$ ) spin state in a 1:3 ratio.<sup>[82,90,91]</sup> The direct transition from  $^3CT$  to the ground state is spin-forbidden, but the triplet CT state may undergo back electron transfer to triplet excitons ( $T_1$ ). Hence, there would be in principle two different decay channels with different relaxation kinetics, which would make the relation between generation and recombination less straightforward.<sup>[82]</sup> For back electron transfer to be relevant, the  $T_1$  level in either the donor or acceptor must be at lower energy than the CT state. Even though such a configuration is not typical for P3HT:PCBM, it cannot be completely ruled out from this work. However, it could be shown that in other materials the loss

channel through triplets is turned off upon aggregation.<sup>[90,91]</sup> Thus, if triplets were relevant, it is likely that they would lead to additional losses in the CB rather than in the DCB system, which enforces our view that aggregation is key to suppress charge recombination. Clarification of this aspect would be an interesting direction for future research.

Summarizing, we find that P3HT:PCBM blends processed via the DCB route display an optimal morphology in terms of reduced recombination and thickness-insensitive device performance. The optimal morphology consists of both amorphous and aggregated regions. To this end, our work confirms earlier suggestions that a three-phase morphology balances best between efficient generation (mainly to occur in the amorphous phase) and reduced recombination (carriers are pushed away from the interface toward aggregated regions), and outperforms both purely amorphous and highly ordered blends.<sup>[20,22]</sup> However, the crucial point here is that the mere existence of aggregates in an amorphous matrix is not sufficient to suppress recombination to such an extent that efficient thick-film devices are possible. To achieve this, the aggregates must be of high crystalline quality and purity, which in the present case is only realized in the carefully equilibrated DCB blends. Since the morphological features discussed here are not exceptional but typical for most OPV blends, we expect our results to be transferable to other systems. In particular, we expect that the guiding principle of having high-quality aggregates is also valid for the acceptor phase, which is supported by a number of recent studies on both fullerenes and NFAs.<sup>[21,23–25]</sup>

## 4. Conclusions

In this paper, we revisited the classical P3HT:PCBM blend to establish connections between the nanoscale morphology and device physics of organic bulk-heterojunction solar cells. By exploiting a structure–property relationship that has so far received little attention, we could show that aggregation is the key feature to reduce recombination losses. However, in order to reduce the recombination rate to such an extent that the solar cells operate as Shockley-type devices even in thick junctions, the mere presence of aggregates is not sufficient. For this to be the case, the aggregates must be of high crystalline quality and purity, so that the charge carriers are delocalized over larger areas. The delocalization boosts the dissociation of charger-transfer states that are formed by the encounter of free electrons and holes. In the case of P3HT:PCBM, such a situation is realized in carefully equilibrated blend films that are slowly dried after spin-coating. The optimized blends show extraordinarily long hole diffusions lengths exceeding 100 nm and can also tolerate the build-up of space charge due to imbalanced transport.

In contrast to the order and energy landscape within the phases, we find that the structure size of the phase separation plays only a minor role in charge recombination. The fact that charge carriers are confined in donor and acceptor domains, so that a coarser phase separation would be preferable from a geometric point of view, is far outperformed by the effect of aggregation. Therefore, optimization of the domain size, for example through nanostructuring, is not a promising approach to

significantly suppress recombination. Instead, the focus should lie on molecular order, crystalline quality, and phase purity. This is especially relevant for devices based on nonfullerene acceptors, where a reduction of the recombination rate is crucial due to the typically low carrier mobilities.

## 5. Experimental Section

**Materials:** Regioregular P3HT was purchased from Rieke Metals (4002-E, molecular weight 50–70 kDa, regioregularity 91–94%). PCBM was purchased from Solenne BV (purity 99.5%). Chlorobenzene (CB), 1,2-dichlorobenzene (DCB) and polyethylenimine (PEIE) were purchased from Sigma-Aldrich. Poly(3,4-ethylenedioxythiophene) polystyrene sulfonate (PEDOT:PSS) was purchased from Heraeus (Clevios P VP Al 4083). Indium tin oxide (ITO) covered glass substrates were purchased from Präzisions Glas & Optik.

**Blend-Film Preparation:** Blend solutions were prepared by dissolving P3HT and PCBM in 1:1 weight ratio in either CB or DCB and stirred at 60 °C for 12 h prior to further processing. Blend films were produced by dynamic spin coating. The thickness was controlled by the concentration of the solution (30 to 60 mg mL<sup>-1</sup>) and the spin-coating speed (500 to 1500 rpm). After complete drying in a closed vessel, all samples were thermally annealed at 150 °C for 10 min. All preparation was carried out under a dry nitrogen atmosphere.

**Electron Tomography:** Specimens for TEM were prepared by depositing blend films on glass substrates covered with a sacrificial layer of PEDOT:PSS. After immersion in deionized water, the PEDOT:PSS layer dissolved, and the free-floating blends were transferred to 300-mesh copper grids. Bright-field TEM images were acquired with a 200-kV field emission electron microscope (Jeol JEM-2100F) at an underfocus of 10 µm.<sup>[42,43,45]</sup> Electron tomography was performed by recording a series of TEM images under different viewing angles by tilting the specimen in a range of ±65° in nonequidistant increments according to Saxton et al.<sup>[92]</sup>. The tilt series was assembled to a volumetric reconstruction (voxel size 0.43 nm) using the simultaneous iterative reconstruction technique with 25 iterations. Acquisition, alignment, and reconstruction of tomographic data was done with the software package Tomography (System in Frontier, Inc.). For the FFT analysis, the reconstructed volumes were cut into series of horizontal slices. Binarization of the grayscale slices was done by applying a median filter (9 × 9 pixels) and thresholding using Otsu's method.

**Device Fabrication:** Inverted solar cells were fabricated by spin coating a 50-nm layer of ZnO nanoparticles (diameter 5 nm, see ref. [93] for details) on cleaned and patterned ITO substrates. Subsequently, the active layer was deposited either from CB or DCB as described above. After thermal annealing, a MoO<sub>3</sub> (12 nm)/Ag (150 nm) electrode was evaporated under high vacuum (10<sup>-6</sup> mbar). The active area was 0.3 cm<sup>2</sup>. Solar cells were encapsulated with glass slides and an UV-cured optical adhesive. Single-carrier diodes were fabricated with the device architecture ITO/PEIE/P3HT:PCBM/Ca/Al (electrons) and ITO/PEDOT:PSS/P3HT:PCBM/MoO<sub>3</sub>/Ag (holes), respectively.

**Characterizations:** Current–voltage curves were recorded with a parameter analyzer (Keithley 4200). A class AAA solar simulator (Photo Emission Tech) was used to provide simulated AM1.5G illumination at 100 mW cm<sup>-2</sup>. The EQE was measured with a custom-built setup (Bentham PVE300), equipped with a 75-W Xe arc lamp and a monochromator. Photocurrent signals were modulated at 780 Hz and monitored with a lock-in amplifier (Stanford Research Systems SR830). White-light bias illumination was provided by a 50-W halogen lamp. The intensity of all light sources used was calibrated with a KG5-filtered reference solar cell. UV–vis absorption spectra were recorded from optically thin films on glass with a spectrophotometer (Varian Cary 100) and corrected for the transmission of the substrate. The P3HT absorption component was fitted to the Spano model,<sup>[54–56]</sup> which treats the absorption spectrum as a series of Gaussian bands,

$$A \propto \sum_{m=0}^{\infty} \left( \frac{S^m}{m!} \right) \left( 1 - \frac{W e^{-S}}{2E_p} \sum_{n \neq m} \frac{S^n}{n!(n-m)} \right)^2 \times \exp \left( - \frac{\left( \hbar\omega - E_{0-0} - mE_p - \frac{1}{2} \frac{WS^m e^{-S}}{m!} \right)^2}{2\sigma^2} \right) \quad (9)$$

where  $\hbar\omega$  is the photon energy,  $S$  the Huang–Rhys factor,  $E_p$  the intramolecular vibrational energy,  $E_{0-0}$  the energy of the 0–0 transition,  $\sigma$  the Gaussian bandwidth, and  $W$  the intramolecular free exciton bandwidth. Assuming  $S = 1$  and  $E_p = 0.179$  eV for the C=C symmetric stretch,<sup>[55]</sup> the only free fit parameters were  $E_{0-0}$ ,  $\sigma$  and  $W$ . Film thicknesses were measured with a stylus profiler (Veeco Dektak 6M).

**Transient Measurements:** For TPC and TPV measurements, a 4-W white-light LED (Seoul P4) was used to provide constant background illumination. Another LED (wavelength 525 nm, 250 ns rise/fall time), driven by a double pulse generator (Agilent 81150A), was used to apply a small optical perturbation to the sample. The second channel of the pulse generator served as bias-voltage source. Current and voltage transients were recorded with a 1-GHz digital storage oscilloscope (Tektronix DPO7104) at an input impedance of 50 Ω and 1 MΩ, respectively. The TPC transients were routinely corrected for RC time effects as described elsewhere.<sup>[94]</sup> A biased silicon detector (Thorlabs DET36A) was used to monitor the switching dynamics and background light intensity. The latter was pre-adjusted for each device by matching the current–voltage response under simulated sunlight. Light sources were attenuated with neutral density filters. Experiments were done at room temperature and ambient pressure.

**Transfer-Matrix Model:** 1D transfer-matrix calculations were performed with a customized MATLAB code.<sup>[95]</sup> The optical constants of all materials involved were determined by spectroscopic ellipsometry. The validity of the model was checked by comparing simulated and experimental reflectance spectra of complete OPV devices. Further details are given in the Supporting Information.

## Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

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## Conflict of Interest

The authors declare no conflict of interest.



## Keywords

aggregation, charge recombination, morphology, organic photovoltaics, phase separation

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- [1] Y. Firdaus, V. M. Le Corre, J. I. Khan, Z. Kan, F. Laquai, P. M. Beaujuge, T. D. Anthopoulos, *Adv. Sci.* **2019**, 6, 1802028.
- [2] Y. Cui, H. Yao, J. Zhang, K. Xian, T. Zhang, L. Hong, Y. Wang, Y. Xu, K. Ma, C. An, C. He, Z. Wei, F. Gao, J. Hou, *Adv. Mater.* **2020**, 32, 1908205.
- [3] Y. Lin, Y. Firdaus, F. H. Isikgor, M. I. Nugraha, E. Yengel, G. T. Harrison, R. Hallani, A. El-Labban, H. Faber, C. Ma, X. Zheng, A. Subbiah, C. T. Howells, O. M. Bakr, I. McCulloch, S. D. Wolf, L. Tsetseris, T. D. Anthopoulos, *ACS Energy Lett.* **2020**, 5, 2935.
- [4] A. Armin, Z. Chen, Y. Jin, K. Zhang, F. Huang, S. Shoaee, *Adv. Energy Mater.* **2018**, 8, 1701450.
- [5] C. Sun, F. Pan, H. Bin, J. Zhang, L. Xue, B. Qiu, Z. Wei, Z. G. Zhang, Y. Li, *Nat. Commun.* **2018**, 9, 743.
- [6] Y. Jin, Z. Chen, M. Xiao, J. Peng, B. Fan, L. Ying, G. Zhang, X. F. Jiang, Q. Yin, Z. Liang, F. Huang, Y. Cao, *Adv. Energy Mater.* **2017**, 7, 1700944.
- [7] P. Meredith, A. Armin, *Nat. Commun.* **2018**, 9, 5261.
- [8] T. M. Clarke, G. Zhang, A. J. Mozer, in *Advanced Concepts in Photovoltaics* (Eds: A. J. Nozik, C. G., M. C. Beard), The Royal Society of Chemistry, Cambridge **2014**.
- [9] D. Bartsaghi, I. Del Carmen Pérez, J. Kniepert, S. Roland, M. Turbiez, D. Neher, L. J. A. Koster, *Nat. Commun.* **2015**, 6, 7083.
- [10] P. Kaienburg, U. Rau, T. Kirchartz, *Phys. Rev. Appl.* **2016**, 6, 024001.
- [11] D. Neher, J. Kniepert, A. Elimelech, L. J. A. Koster, *Sci. Rep.* **2016**, 6, 24861.
- [12] C. Göhler, A. Wagenpfahl, C. Deibel, *Adv. Electron. Mater.* **2018**, 4, 1700505.
- [13] G. Lakhwani, A. Rao, R. H. Friend, *Annu. Rev. Phys. Chem.* **2014**, 65, 557.
- [14] A. Pivrikas, G. Juška, A. J. Mozer, M. Scharber, K. Arlauskas, N. S. Saricifci, H. Stubb, R. Österbacka, *Phys. Rev. Lett.* **2005**, 94, 176806.
- [15] L. J. A. Koster, V. D. Mihailetschi, P. W. M. Blom, *Appl. Phys. Lett.* **2006**, 88, 052104.
- [16] C. Groves, N. C. Greenham, *Phys. Rev. B* **2008**, 78, 155205.
- [17] M. C. Heiber, C. Baumbach, V. Dyakonov, C. Deibel, *Phys. Rev. Lett.* **2015**, 114, 136602.
- [18] V. Coropceanu, J. L. Brédas, S. Mehraeen, *J. Phys. Chem. C* **2017**, 121, 24954.
- [19] T. M. Burke, S. Sweetnam, K. Vandewal, M. D. McGehee, *Adv. Energy Mater.* **2015**, 5, 1500123.
- [20] M. Nyman, O. J. Sandberg, R. Österbacka, *Adv. Energy Mater.* **2015**, 5, 1400890.
- [21] H. Cha, G. Fish, J. Luke, A. Alraddadi, H. H. Lee, W. Zhang, Y. Dong, S. Limbu, A. Wadsworth, I. P. Maria, L. Francás, H. L. Sou, T. Du, J. Kim, M. A. McLachlan, I. McCulloch, J. R. Durrant, *Adv. Energy Mater.* **2019**, 9, 1901254.
- [22] K. N. Schwarz, P. B. Geraghty, V. D. Mitchell, S.-U.-Z. Khan, O. J. Sandberg, N. Zarrabi, B. Kudisch, J. Subbiah, T. A. Smith, B. P. Rand, A. Armin, G. D. Scholes, D. J. Jones, K. P. Ghiggino, *J. Am. Chem. Soc.* **2020**, 142, 2562.
- [23] S. Wilken, T. Upreti, A. Melianas, S. Dahlström, G. Persson, E. Olsson, R. Österbacka, M. Kemerink, *Sol. RRL* **2020**, 4, 2000029.
- [24] L. Liu, Y. Kan, K. Gao, J. Wang, M. Zhao, H. Chen, C. Zhao, T. Jiu, A.-K.-Y. Jen, Y. Li, *Adv. Mater.* **2020**, 32, 1907604.
- [25] S. M. Hosseini, N. Tokmoldin, Y. W. Lee, Y. Zou, H. Y. Woo, D. Neher, S. Shoaee, *Sol. RRL* **2020**, 4, 202000498.
- [26] A. J. Ferguson, N. Kopidakis, S. E. Shaheen, G. Rumbles, *J. Phys. Chem. C* **2011**, 115, 23134.
- [27] N. Gasparini, M. Salvador, T. Heumüller, M. Richter, A. Classen, S. Shrestha, G. J. Matt, S. Holliday, S. Strohm, H. J. Egelhaaf, A. Wadsworth, D. Baran, I. McCulloch, C. J. Brabec, *Adv. Energy Mater.* **2017**, 7, 1701561.
- [28] J. I. Khan, R. S. Ashraf, M. A. Alamoudi, M. N. Nabi, H. N. Mohammed, A. Wadsworth, Y. Firdaus, W. Zhang, T. D. Anthopoulos, I. McCulloch, F. Laquai, *Solar RRL* **2019**, 3, 1900023.
- [29] S. Holliday, R. S. Ashraf, A. Wadsworth, D. Baran, S. A. Yousaf, C. B. Nielsen, C.-H. Tan, S. D. Dimitrov, Z. Shang, N. Gasparini, M. Alamoudi, F. Laquai, C. J. Brabec, A. Salleo, J. R. Durrant, I. McCulloch, *Nat. Commun.* **2016**, 7, 11585.
- [30] D. Baran, R. S. Ashraf, D. A. Hanifi, M. Abdelsamie, N. Gasparini, J. A. Röhr, S. Holliday, A. Wadsworth, S. Lockett, M. Neophytou, J. M. Emmott, J. Nelson, C. J. Brabec, A. Amassian, A. Salleo, T. Kirchartz, J. R. Durrant, I. McCulloch, *Nat. Mater.* **2017**, 16, 363.
- [31] B. Xiao, A. Tang, J. Zhang, A. Mahmood, Z. Wei, E. Zhou, *Adv. Energy Mater.* **2017**, 7, 10168.
- [32] Y. Wu, H. Bai, Z. Wang, P. Cheng, S. Zhu, Y. Wang, W. Ma, X. Zhan, *Energy Environ. Sci.* **2015**, 8, 3215.
- [33] J. Wu, Y. Xu, Z. Yang, Y. Chen, X. Sui, L. Yang, P. Ye, T. Zhu, X. Wu, X. Liu, H. Cao, A. Peng, H. Huang, *Adv. Energy Mater.* **2019**, 9, 1803012.
- [34] C. Yang, S. Zhang, J. Ren, M. Gao, P. Bi, L. Ye, J. Hou, *Energy Environ. Sci.* **2020**, 13, 2864.
- [35] J. Kniepert, I. Lange, N. J. van der Kaap, L. J. A. Koster, D. Neher, *Adv. Energy Mater.* **2014**, 4, 1301401.
- [36] C. Deibel, A. Baumann, V. Dyakonov, *Appl. Phys. Lett.* **2008**, 93, 163303.
- [37] G. Li, V. Shrotriya, J. Huang, Y. Yao, T. Moriarty, K. Emery, Y. Yang, *Nat. Mater.* **2005**, 4, 864.
- [38] S. Wilken, O. J. Sandberg, D. Scheunemann, R. Österbacka, *Sol. RRL* **2020**, 4, 1900505.
- [39] S. D. Oosterhout, M. M. Wienk, S. S. van Bavel, R. Thiedmann, L. J. A. Koster, J. Gilot, L. J., V. Schmidt, R. A. J. Janssen, *Nat. Mater.* **2009**, 8, 818.
- [40] S. S. van Bavel, E. Sourty, G. de With, J. Loos, *Macromolecules* **2009**, 42, 7396.
- [41] S. T. Turner, P. Pingel, R. Steyrleuthner, E. J. W. Crossland, S. Ludwigs, D. Neher, *Adv. Funct. Mater.* **2011**, 21, 4640.
- [42] W. Ma, C. Yang, A. J. Heeger, *Adv. Mater.* **2007**, 19, 1387.
- [43] J. S. Moon, J. K. Lee, S. Cho, J. Byun, A. J. Heeger, *Nano Lett.* **2009**, 9, 230.
- [44] S. S. van Bavel, E. Sourty, G. de With, J. Loos, *Nano Lett.* **2009**, 9, 507.
- [45] P. Dutta, Y. Xie, M. Kumar, M. Rath, P. Ahrenkiel, D. Galipeau, Q. Qiao, V. BommiSETTY, *J. Photonics Energy* **2011**, 1, 011124.
- [46] M. Campoy-Quiles, T. Ferenczi, T. Agostinelli, P. G. Etchegoin, Y. Kim, T. D. Anthopoulos, P. N. Stavrinou, D. D. C. Bradley, J. Nelson, *Nat. Mater.* **2008**, 7, 158.
- [47] D. Legland, K. Kiêu, M.-F. Devaux, *Image Anal. Stereol.* **2007**, 26, 83.
- [48] C.-W. Chu, H. Yang, W.-J. Hou, J. Huang, G. Li, Y. Yang, *Appl. Phys. Lett.* **2008**, 92, 103306.
- [49] Y. Xie, Y. Bao, J. Du, C. Jiang, Q. Qiao, *Phys. Chem. Chem. Phys.* **2012**, 14, 10168.
- [50] P. J. Brown, D. S. Thomas, A. Köhler, J. S. Wilson, J.-S. Kim, C. M. Ramsdale, H. Sirringhaus, R. H. Friend, *Phys. Rev. B* **2003**, 67, 064203.
- [51] K. Tremel, S. Ludwigs, *Adv. Polym. Sci.* **2014**, 265, 39.

- [52] H. Sirringhaus, P. J. Brown, R. H. Friend, M. M. Nielsen, K. Bechgaard, B. M. W. Langeveld-Voss, A. J. H. Spiering, R. A. J. Janssen, E. W. Meijer, P. Herwig, D. M. de Leeuw, *Nature* **1999**, 401, 685.
- [53] A. J. Moulé, K. Meerholz, *Adv. Mater.* **2008**, 20, 240.
- [54] F. C. Spano, *J. Chem. Phys.* **2005**, 122, 234701.
- [55] J. Clark, C. Silva, R. H. Friend, F. C. Spano, *Phys. Rev. Lett.* **2007**, 98, 206406.
- [56] J. Clark, J. F. Chang, F. C. Spano, R. H. Friend, C. Silva, *Appl. Phys. Lett.* **2009**, 94, 163306.
- [57] D. Beljonne, J. Cornil, R. Silbey, P. Millié, J. L. Brédas, *J. Chem. Phys.* **2000**, 112, 4749.
- [58] J. Gierschner, Y. S. Huang, B. van Averbeke, J. Cornil, R. H. Friend, D. Beljonne, *J. Chem. Phys.* **2009**, 130, 044105.
- [59] P. Pingel, A. Zen, R. D. Abellón, F. C. Grozema, L. D. A. Siebbeles, D. Neher, *Adv. Funct. Mater.* **2010**, 20, 2286.
- [60] J. Patel, A. Sharma, M. Chauhan, M. Aatif, N. Vashistha, M. Kumar, B. Tripathi, S. Chand, J. P. Tiwari, M. K. Pandey, *Phys. Chem. Chem. Phys.* **2019**, 21, 2771.
- [61] P. Kohn, Z. Rong, K. H. Scherer, A. Sepe, M. Sommer, P. Müller-Buschbaum, R. H. Friend, U. Steiner, S. Hüttner, *Macromolecules* **2013**, 46, 4002.
- [62] N. D. Treat, M. A. Brady, G. Smith, M. F. Toney, E. J. Kramer, C. J. Hawker, M. L. Chabiny, *Adv. Energy Mater.* **2011**, 1, 82.
- [63] M. Pfannmöller, H. Flügge, G. Benner, I. Wacker, C. Sommer, M. Hanselmann, S. Schmale, H. Schmidt, F. A. Hamprecht, T. Rabe, W. Kowalsky, R. R. Schröder, *Nano Lett.* **2011**, 11, 3099.
- [64] R. C. Masters, A. J. Pearson, T. S. Glen, F.-C. Sasam, L. Li, M. Dapor, A. M. Donald, D. G. Lidzey, C. Rodenburg, *Nat. Commun.* **2015**, 6, 6928.
- [65] W. F. Pasveer, J. Cottaar, C. Tanase, R. Coehoorn, P. A. Bobbert, P. W. M. Blom, D. M. de Leeuw, M. A. J. Michels, *Phys. Rev. Lett.* **2005**, 94, 206601.
- [66] N. Felekidis, A. Melianas, M. Kemerink, *Org. Electron.* **2018**, 61, 318.
- [67] R. Noriega, J. Rivnay, K. Vandewal, F. P. V. Koch, N. Stingelin, P. Smith, M. F. Toney, A. Salleo, *Nat. Mater.* **2013**, 12, 1038.
- [68] S. Roland, J. Kniepert, J. A. Love, V. Negi, F. Liu, P. Bobbert, A. Melianas, M. Kemerink, A. Hofacker, D. Neher, *J. Phys. Chem. Lett.* **2019**, 10, 1374.
- [69] S. Dahlström, C. Ahläng, K. Björkström, S. Forsblom, B. Granroth, K. Jansson, A. Luukkonen, M. T. Masood, J. Poulizac, S. Qudsia, M. Nyman, *AIP Adv.* **2020**, 10, 065203.
- [70] V. D. Mihailetschi, H. Xie, B. De Boer, L. J. A. Koster, P. W. M. Blom, *Adv. Funct. Mater.* **2006**, 16, 699.
- [71] J. A. Bartelt, D. Lam, T. M. Burke, S. M. Sweetnam, M. D. McGehee, *Adv. Energy Mater.* **2015**, 5, 1500577.
- [72] S. R. Cowan, R. A. Street, S. Cho, A. J. Heeger, *Phys. Rev. B* **2011**, 83, 035205.
- [73] Z. Li, F. Gao, N. C. Greenham, C. R. McNeill, *Adv. Funct. Mater.* **2011**, 21, 1419.
- [74] U. Würfel, L. Perdígón-Toro, J. Kurpiers, C. M. Wolff, P. Caprioglio, J. J. Rech, J. Zhu, X. Zhan, W. You, S. Shoaee, D. Neher, M. Stolterfoht, *J. Phys. Chem. Lett.* **2019**, 10, 3473.
- [75] A. Foertig, J. Rauh, V. Dyakonov, C. Deibel, *Phys. Rev. B* **2012**, 86, 115302.
- [76] T. Kirchartz, J. Nelson, *Phys. Rev. B* **2012**, 86, 165201.
- [77] T. Kirchartz, B. E. Pieters, J. Kirkpatrick, U. Rau, J. Nelson, *Phys. Rev. B* **2011**, 83, 115209.
- [78] D. Kiermasch, A. Baumann, M. Fischer, V. Dyakonov, K. Tvingstedt, *Energy Environ. Sci.* **2018**, 11, 629.
- [79] O. J. Sandberg, K. Tvingstedt, P. Meredith, A. Armin, *J. Phys. Chem. C* **2019**, 123, 14261.
- [80] M. C. Heiber, T. Okubo, S.-J. Ko, B. R. Luginbuhl, N. A. Ran, M. Wang, H. Wang, M. A. Uddin, H. Y. Woo, G. C. Bazan, T.-Q. Nguyen, *Energy Environ. Sci.* **2018**, 11, 3019.
- [81] N. Tokmoldin, S. M. Hosseini, M. Raoufi, L. Q. Phuong, O. J. Sandberg, H. Guan, Y. Zou, D. Neher, S. Shoaee, *J. Mater. Chem. A* **2020**, 8, 7854.
- [82] S. Shoaee, A. Armin, M. Stolterfoht, S. M. Hosseini, J. Kurpiers, D. Neher, *Sol. RRL* **2019**, 3, 1900184.
- [83] D. H. K. Murthy, A. Melianas, Z. Tang, G. Juška, K. Arlauskas, F. Zhang, L. D. A. Siebbeles, O. Inganäs, T. J. Savenije, *Adv. Funct. Mater.* **2013**, 23, 4262.
- [84] C. Deibel, T. Strobel, V. Dyakonov, *Adv. Mater.* **2010**, 22, 4097.
- [85] K. Tvingstedt, K. Vandewal, A. Gadisa, F. Zhang, J. Manca, O. Inganäs, *J. Am. Chem. Soc.* **2009**, 131, 11819.
- [86] J. Gorenflot, M. C. Heiber, A. Baumann, J. Lorrman, M. Gunz, A. Kämpgen, V. Dyakonov, C. Deibel, *J. Appl. Phys.* **2014**, 115, 144502.
- [87] F. C. Jamieson, E. B. Domingo, T. McCarthy-Ward, M. Heeney, N. Stingelin, J. R. Durrant, *Chem. Sci.* **2012**, 3, 485.
- [88] S. Sweetnam, K. R. Graham, G. O. Ngongang Ndjawa, T. Heumüller, J. A. Bartelt, T. M. Burke, W. Li, W. You, A. Amassian, M. D. McGehee, *J. Am. Chem. Soc.* **2014**, 136, 14078.
- [89] R. Österbacka, C. P. An, X. M. Jiang, Z. V. Vardeny, *Science* **2000**, 287, 839.
- [90] A. Rao, P. C. Y. Chow, S. Gélinas, C. W. Schlenker, C. Z. Li, H. L. Yip, A. K. Y. Jen, D. S. Ginger, R. H. Friend, *Nature* **2013**, 500, 435.
- [91] P. C. Y. Chow, S. Gélinas, A. Rao, R. H. Friend, *J. Am. Chem. Soc.* **2014**, 136, 3424.
- [92] W. O. Saxton, W. Baumeister, M. Hahn, *Ultramicroscopy* **1984**, 13, 57.
- [93] S. Wilken, J. Parisi, H. Borchert, *J. Phys. Chem. C* **2014**, 118, 19672.
- [94] S. W. Kettlitz, J. Mescher, N. S. Christ, M. Nintz, S. Valouch, A. Colsmann, U. Lemmer, *IEEE Photonics Technol. Lett.* **2013**, 25, 682.
- [95] G. F. Burkhard, E. T. Hoke, M. D. McGehee, *Adv. Mater.* **2010**, 22, 3293.