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1 **From wood pulp fibers to tubular SiO₂/C composite as anode for Li-ion battery:**
2 **in-situ regulation of cellulose microfibrils by alkali solution**

3 Xiaodi Wang ^a, Xiuzhi Liu ^a, Xiaoli Ren ^a, Kaisheng Luo ^a, Wenyang Xu ^c, Qingxi Hou ^{a,b*}, Wei
4 Liu ^{a*}

5
6
7 ^a Tianjin Key Laboratory of Pulp and Paper, Tianjin University of Science & Technology, Tianjin
8 300457, China

9 ^b State Key Laboratory of Pulp and Paper Engineering, South China University of Technology,
10 Guangzhou 510640, China

11 ^c Johan Gadolin Process Chemistry Centre, Laboratory of Natural Materials Technology, Åbo
12 Akademi University, Turku FI-20500, Finland

13 **Corresponding authors**

14 * Tianjin Key Laboratory of Pulp and Paper, Tianjin University of Science & Technology,
15 Tianjin 300457, China. Email: qingxihou@tust.edu.cn (Q. X. Hou); wei.liu2009@hotmail.com
16 (W. Liu)

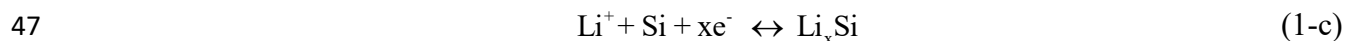
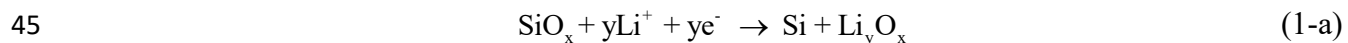
17 **Abstract**

18 The microfibrils of wood pulp fibers in plant cell wall with an intrinsically regular and interlaced
19 arrangement offer a stable porous structure. They could also provide outstanding mechanical
20 strength and promote conductivity for Si-based materials. Upon these benefits, a method using
21 wood pulp fibers to fabricate SiO₂/C composite is proposed. In this work, the alkali pretreatment
22 interfering hydrogen bonds in wood pulp fibers *in situ* results in an enlarged spatial distance among
23 cellulose microfibrils, exposing more free-surface. This empowered homogenously inlaid SiO₂
24 distribution along the carbon skeleton in the subsequent carbonization process. A highly
25 mesoporous and the tubular morphology was achieved in the fabricated SiO₂/C composite.
26 Furthermore, a high initial coulombic efficiency of 84.9% and reversible specific capacity of 1130
27 mAh g⁻¹ with a coulombic efficiency of 98.8% after 200 cycles at a current density of 0.1 A g⁻¹
28 was obtained.

29 Keywords: Cellulose, Wood pulp fibers, SiO₂/C composite, Alkali pretreatment

30 1. Introduction

31 Lithium-ion batteries (LIBs) with characteristics of high energy density and environmental
32 friendliness have attracted growing attention for electric energy storage in versatile fields such as
33 mobile electronics and electric vehicles (Chen et al., 2017b; Shi et al., 2017; Tao et al., 2017).
34 Abundant materials have been explored to replace graphite as the anode materials for LIBs to
35 endow LIBs with a high specific capacity and cycle performance (Roy et al., 2016). Within the
36 family of anode materials, Si-based materials, *i.e.*, SiO_x (x = 0, 1, 2), become a promising candidate
37 mainly due to their high theoretical specific capacity. The coexistence of the irreversible and
38 reversible lithiation mechanisms is well established in SiO_x-based anode (Equation 1a-c) (Favors
39 et al., 2014). It allows a favorable achievement of a better cycling performance of the SiO₂-based
40 anode than that of the pure Si-based anode due to the *in situ* formation of Li₄O₂ and Li₄SiO₄ serving
41 as buffer media to suppress volume change (Chang et al., 2012; Zeng et al., 2018). Moreover, the
42 inherent detriments of Si-based materials, such as poor electronic conductivity and low ion
43 diffusivity, hamper the application of SiO₂-based anode (Xiao et al., 2019; Xu et al., 2018). SiO₂/C
44 composites were designed to mitigate the abovementioned problems (Luo et al., 2015).



48 Wood pulp fibers, one type of abundant and sustainable natural carbon-rich resources, supply
49 a promising bio-template in the fabrication of energy storage composites (Tonoli et al., 2010).
50 Gao's group (Yang et al., 2014) applied wood pulp fibers to assemble a novel MnO/C composite

51 possessing a high discharge capacity, stable cycle stability, and excellent rate performance. This is
52 because the mesoporous structures of wood pulp fibers in plant cell wall was well preserved after
53 carbonization (Luo et al., 2017). Additionally, it could be served as ion diffusion channels of the
54 designate electrochemical energy storage materials during the charge and discharge processes.
55 Similarly, Kim et al. proved (Kim et al., 2018b) that an improved electrochemical performance was
56 achieved by utilizing naturally hierarchical mesoporous structure of wood pulp fibers functioning
57 as electron paths. The macro- and micro-pores in carbon fibers originated from wood fibers had
58 insufficient space for metal oxide to alleviate its volume expansion during lithiation and de-
59 lithiation processes (Zhang et al., 2020). In addition, the unique micro-nano structure of porous
60 carbon derived from wood pulp fibers could not only reduce the first irreversible capacity loss by
61 limiting the contact between the electrode and electrolyte, but also enhance the specific capacity
62 by accelerating the transfer of electrons and ions through nanopores (Zheng et al., 2016). Owing to
63 the amorphous structure, large carbon interlayer, and fibrous morphology with hollow
64 constructions, wood pulp fibers offered the carbon materials with remarkable electrochemical
65 performance with favourable lithium/sodium ion storage and efficient transfer (Zhang et al., 2019).
66 Based on the above, using wood pulp fibers to fabricate SiO₂/C composite is worth considering.

67 However, in order to facilitate storage and transportation, the wood pulp fibers have to be
68 subjected to pressing and drying. This caused the deformation of the fibers, *i.e.*, pore collapse and
69 lumen disappearance (Hribernik et al., 2016; Kimura et al., 2014). It further turns to insufficient
70 space for ion transport as it will be used to fabricate SiO₂/C composite as the current state. In
71 addition, the cell wall of wood pulp fibers is composed of a large number of cellulose molecular

72 chains which are arranged and aggregated sequentially. The aggregation of cellulose molecular
73 chains results in two states: crystalline region and non-crystalline region (*i.e.* amorphous region).
74 Particularly in the crystalline region, cellulose chains are tightly packed at the presence of inter-
75 hydrogen bonds, causing low SiO₂ load from the low absorption rate of the precursor solution. An
76 appropriate pretreatment is highly demanding to interfere the crystalline region and reopen the cell
77 lumen in wood pulp fibers for fabricating SiO₂/C composites as high-performance anode material
78 for LIBs.

79 Alkali treatment is well recognized in modifying the structure and properties of wood pulp
80 fibers (Fink et al., 1999; Ishikura et al., 2010; Nakano, 2010; Zhang et al., 2005), because of its
81 effects on the crystallinity and supermolecular structure of cellulose (Dinand et al., 2002; Eronen
82 et al., 2009; Nie et al., 2018). It opens up a totally new avenue in the preparation of porous carbon
83 scaffolds by offering the increased wrinkles and pores on the surface of the fibers. Moreover, the
84 high SiO₂ load in the SiO₂/C composite can be obtained due to both the decrease in the total
85 crystallinity (*i.e.*, the sum of the degrees of crystallinity for cellulose I and cellulose II) and the
86 increased exposure of the accessible surface of wood pulp fibers after alkali treatment (Chen et al.,
87 2017a; Halonen et al., 2013; Hribernik et al., 2016).

88 In the present work, a novel fabrication method of SiO₂/C composite by *in situ* synthesized
89 method was established. The wood pulp fibers were pretreated with different concentration levels
90 of alkali solution before immersing in the tetraethoxysilane (TEOS) ethanol solution. The exposure
91 of more hydroxyl groups and the tubular morphology of wood pulp fibers can be achieved after
92 alkali pretreatment. This endowed the homogeneous distribution of the silicon element on the

93 surface and inner wall of the tubular SiO₂/C composite after carbonization. Both inner and outer
94 silica layers could participate in the lithiation reaction by using the composite as anode material
95 for LIBs. The capacity of liquid retention for the pretreated pulp fibers and the changes in the
96 crystallinity of cellulose were both investigated. Then the micro-morphology, crystal structure and
97 porous properties of the SiO₂/C composites prepared with the pretreated pulp fibers as carbon
98 skeleton were analyzed. At last, the electrochemical properties of the SiO₂/C composites were
99 further characterized. The present work successfully demonstrated the feasibility of wood pulp
100 fibers by *in situ* regulation using alkali solution as carbon skeleton to prepare the tubular SiO₂/C
101 composites, offering one potential high-performance anode material for LIBs.

102 **2. Experimental**

103 *2.1 Selection, dispersion, and alkali pretreatment of natural lignocellulosic fibers*

104 The natural lignocellulosic fibers used in this work were from commercial bleached *Pinus*
105 *radiata* kraft pulp made in Chile, of which the content of α -cellulose and ISO brightness were
106 89.42% and 85.98%, respectively, and the average fiber length and width were 2.32 mm and 33.1
107 μm , respectively.

108 The equivalent of 200 g oven-dry wood pulp fibers was taken and well dispersed in 20 L
109 deionized (DI) water by using a Valley beater (No. 2505, KRK Co., Japan) at room temperature
110 and atmosphere pressure for 30 min. It was further thickened using an 80-mesh polycotton bag
111 until the dryness reached about 30%, and sealed in a plastic bag and stored at about 4 °C.

112 The wood pulp fibers were separately immersed in 0, 5, 10, 15, and 17.5 wt% NaOH solution

113 at a solid to liquid ratio of 1:100 (g/mL) at room temperature and a stirring speed of 600 rpm for 1
114 h. After the alkali pretreatment, the wood pulp fibers were rigorously washed with DI water until
115 the pH of the filtrate was neutral. The pulp was then thickened once again to a dryness of about
116 30%. It was sealed in a plastic bag and stored at about 4 °C for later use.

117 2.2 Water retention value (WRV) of the wood pulp fibers

118 The WRV of the wood pulp fibers was principally measured according to the standard method
119 of ISO 23714 (2014). The equivalent of about 300 mg oven-dry wood pulp fibers pretreated with
120 NaOH solution was selected and immersed in 300 mL of DI water for 1h at room temperature, then
121 centrifuged at (3000 ± 50) g for (1800 ± 30) s in a centrifuge (D-37520, SIGMA Laborzentrifugen,
122 German). After centrifugation, the WRV was calculated using the following equation (2):

$$123 \quad WRV = m_2 / m_1 - 1 \quad (2)$$

124 where m_2 and m_1 are the weights of the wood pulp fibers before and after oven-drying in g,
125 respectively.

126 2.3 Crystallinity index (CrI) of the wood pulp fibers

127 *CrI* is an index that can reflect the content of crystalline regions and accessibility of the wood
128 pulp fibers. In the present work, an X-ray diffractometer (XRD-6100, Shimadzu Co., Japan) with
129 a Cu target and a scan speed of 0.02 °/min was used to determine the total *CrI* of the wood pulp
130 fibers, which was the sum of crystallinity for cellulose I and cellulose II. The *CrI* for cellulose I
131 and cellulose II was separately estimated using the following equation (3) in the literature (Buschle-
132 Diller and Zeronian, 1992):

$$133 \quad CrI = (I_2 - I_1) / I_2 \quad (3)$$

134 where I_1 is the intensity at the minimum (between $2\theta = 18^\circ$ and 19° for cellulose I and between 2θ
135 $= 13^\circ$ and 15° for cellulose II), and I_2 is the intensity of the crystalline peak at the maximum
136 (between $2\theta = 22^\circ$ and 23° for cellulose I and between $2\theta = 18^\circ$ and 22° for cellulose II).

137 *2.4 Preparation of SiO₂/C composites*

138 The SiO₂/C composites derived from the wood pulp fibers were fabricated based on the
139 combined sol-gel and heat treatment method (Dirican et al., 2015; Jia et al., 2017a; Jia et al., 2017b).
140 The equivalent of 1.00 g oven-dry wood pulp fibers were first dispersed in 200 mL of the absolute
141 ethanol at room temperature with a stirring speed of 800 rpm for 5 min. A fluffy and thin filter cake
142 was made *via* suction filtration and solvent evaporation. The filter cake was further immersed into
143 40.0 mL of the silicon precursor solution (*i.e.*, 1.2 mol L⁻¹ TEOS-ethanol solution) at room
144 temperature and then taken out from the solution followed by drying in air. The silanol-cellulose
145 composite cake was obtained by repeating the above immersing-drying process until the TEOS
146 solution was fully consumed. The finally obtained silanol-cellulose composite cake was then aged
147 in air at room temperature for 12 h to increase the degree of the silanol condensation. After the
148 completion of the aging treatment, the silanol-cellulose composite cake was carbonized to yield
149 the SiO₂/C composite in argon atmosphere at 600 °C and a heating rate 2 °C min⁻¹ for 6 h. The
150 resultant SiO₂/C composites were correspondingly denoted as 0, 5, 10, 15, and 17.5 wt%-SiO₂/C,
151 respectively.

152 *2.5 Characterization of the resultant SiO₂/C composites*

153 The X-ray diffraction (XRD) patterns of the resultant SiO₂/C composites were recorded on an
154 X-ray diffractometer (XRD-6100, Shimadzu Co., Japan). The Raman spectra of the SiO₂/C

155 composites were measured using a thermoelectrically cooled charge-coupled device (CCD)
156 detector (BTC 162E-532H, B&W TEK Optoelectronics Co., Shanghai, China) equipped with a 532
157 nm solid state laser (Power = 4 mW, Changchun New Industries Optoelectronics Technology Co.,
158 China). SEM images, elemental analysis and elemental mapping images were obtained using a
159 scanning electron microscope (JEOL JSM-IT300LV, Tokyo, Japan) equipped with an energy
160 dispersive X-ray spectrometer (EDX) (X-Max 20, Oxford, England). X-ray photoelectron
161 spectroscopy (XPS) (ESCALAB 250Xi, Thermo Scientific, USA) was carried out on 0, 15 and
162 17.5 wt%-SiO₂/C composites, respectively.

163 The surface features and pore structures of 0, 15 and 17.5 wt%-SiO₂/C composites, namely
164 the Brunauer-Emmett-Teller (BET) specific surface area (SSA), pore size distribution based on
165 density functional theory model, and micropore volume based on t-plot method, were all recorded
166 using a SSA and pore size analyzer (ASIQM0000000-6, Quantachrome Instruments, USA).

167 *2.6 Electrochemical measurements of the SiO₂/C composites*

168 To conduct the electrochemical measurement of the SiO₂/C composites, it is necessary to
169 fabricate the working electrodes. The as-prepared SiO₂/C composite was mixed together with both
170 acetylene black acting as a conductive agent and polyvinylidene fluoride (PVDF) serving as a
171 binder at a weight ratio of 70:20:10. The obtained uniform slurry was then coated onto pure copper
172 foils with a thickness of about 10 μm and dried in a vacuum oven at 120 °C for 24 h. The loading
173 mass of the active material, *i.e.*, the SiO₂/C composite, was about 1.9 mg cm⁻². The dried copper
174 foils with the active material coating were cut into small disks with a diameter of 16 mm. The
175 lithium metal disks with the same diameter were used as both the counter and reference electrodes.

176 The electrolyte was made by dissolving 1.0 mol L⁻¹ lithium hexafluorophosphate (LiPF₆) in the
177 mixture of ethylene carbonate (EC), dimethyl carbonate (DMC) and ethyl methyl carbonate (EMC)
178 at a volume ratio of 1:1:1. A Celgard 2400 porous membrane was used as the separator. At last,
179 CR2032 coin-type cells were assembled in a gloves box (Lab2000, Etelux Co., China) filled with
180 high-purity Argon (H₂O < 0.01 ppm, O₂ < 0.01 ppm).

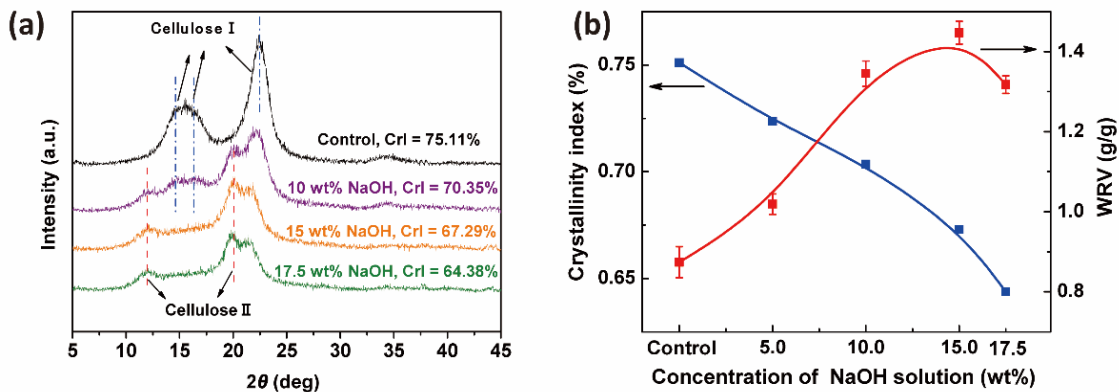
181 Cyclic voltammogram (CV) measurements were carried out using a CHI660D
182 electrochemical workstation (CH instruments Co., China) in a voltage range of 0.01–3.0 V at a
183 scanning rate of 1 mV s⁻¹. Electrochemical impedance spectroscopy (EIS) in a frequency range
184 from 0.1 MHz to 0.01 Hz at an amplitude ratio of 5 mV was also performed using the same
185 CHI660D electrochemical workstation. Galvanostatic charge and discharge measurements were
186 conducted by using a LANHE CT2001A electrochemical workstation (Wuhan Land Co., China) in
187 the voltage range of 0.01–3.0 V at a current density of 0.1 A g⁻¹.

188 **3. Results and discussion**

189 *3.1. Morphologies and accessibility of the alkali-swollen wood pulp fibers*

190 The changes in the cellulosic crystalline structure of the wood pulp fibers after alkaline
191 swelling at different levels of alkali concentration were recorded by XRD patterns, as shown in Fig.
192 1a. It can be found that the pattern of the control wood pulp fibers exhibited a sharp high peak at
193 $2\theta = 22.7^\circ$ and two overlapped weaker diffraction peaks at $2\theta = 15.0^\circ$ and 16.3° , respectively,
194 which are assigned to cellulose I (Liu and Hu, 2008). For the sample treated with 10 wt% NaOH
195 solution, however, two additional distinctive diffraction peaks appeared at the position of $2\theta = 12.0^\circ$

196 and $2\theta = 20.0^\circ$, which are assigned to cellulose II, one kind of allomorph of cellulose I. Cellulose
 197 I gradually transformed into cellulose II in the application of alkali treatment. As is known,
 198 cellulose II has large accessible crystal surface as compared to cellulose I. Therefore, alkali
 199 pretreatment could enhance the accessibility and reactivity of the wood pulp fibers through
 200 changing the crystal structure of the cellulose (Ciolacu et al., 2012). Meanwhile, the total
 201 crystallinity of the cellulose distinctly decreased with raising the concentration of the NaOH
 202 solution used in the alkali pretreatment, as shown in Fig. 1b. The changes in the surface morphology
 203 of the wood pulp fibers were captured by using SEM and depicted in Fig. S1. The alkaline swelling
 204 rendered the morphological changes of the wood pulp fibers from ribbon-like configuration to rod-
 205 like one with coarse surface at the presence of wrinkles and granulations. The resultant increase in
 206 the superficial area accordingly enhanced the accessibility and reactivity of the wood pulp fibers.



207
 208 **Fig. 1.** (a) XRD patterns of wood pulp fibers pretreated with NaOH solution at different levels of concentration,
 209 (b) Effects of alkali pretreatment on the total crystallinity and WRV of wood pulp fibers.

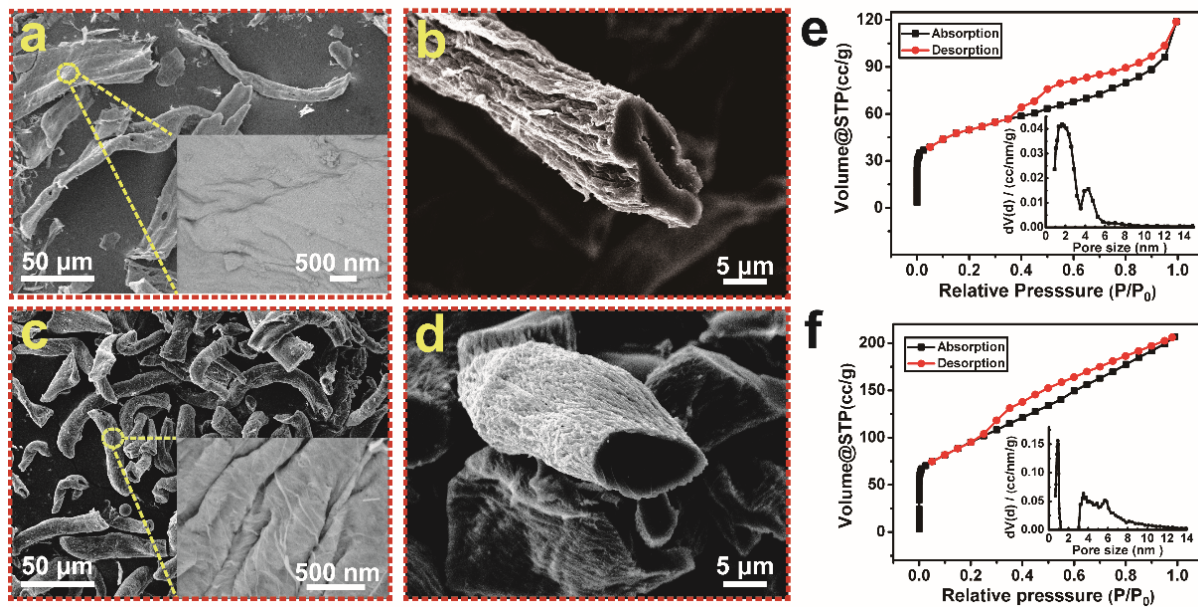
210 Fig. 1b also shows that the higher the concentration of alkali solution, the higher the WRV of
 211 wood pulp fibers, except for the sample pretreated with 17.5 wt% NaOH solution. The alkaline

212 swelling on the wood pulp fibers brought higher cellulose accessibility both in amorphous and
213 crystalline regions than water swelling, which was manifested by the improvement in the WRV of
214 wood pulp fibers. However, an excessively high concentration of NaOH solution can cause a
215 decrease in the number of water molecules introduced into cellulose by hydrated alkali ions or
216 hydrated dipole ions, resulting in the weakening of the swelling effect of alkali solution on cellulose.
217 It is apparent from the above that the concentration of alkali solution played a prominent role in
218 the alkaline swelling of wood pulp fibers. It is of vital importance for wood pulp fibers to have a
219 higher WRV in the process of preparing SiO₂/C composites by using the method of combining sol-
220 gel and heating. This is because there would be more free hydroxyl groups are exposed in the
221 swollen cellulose chains, which is very much beneficial to the retention and uniform distribution
222 of SiO₂ in the target SiO₂/C composite.

223 *3.2. Morphologies and structures of the SiO₂/C composites*

224 The 15 wt%-SiO₂/C composite prepared from the NaOH-swollen wood pulp fibers still
225 maintained the same rod-like structure as that before carbonization (Fig. 2a), and the 0 wt%-SiO₂/C
226 composite derived from untreated wood pulp fibers showed as ribbon-like (Fig. 2c). It means all
227 SiO₂/C composites would conserve their original morphology after carbonization. It can be
228 concluded that alkali treatment is effective to regulate the morphology of SiO₂/C composites. The
229 cross-sectional images of the corresponding SiO₂/C composites are shown in Fig. 2b and d,
230 respectively, indicating that the larger lumen was only found in 15 wt%-SiO₂/C composite. Overall,
231 15 wt%-SiO₂/C composite possessed tubular morphology, which is preferred to make lithium
232 diffusion distance shorter and to have enough space to prevent the pulverization of silicon (Su et

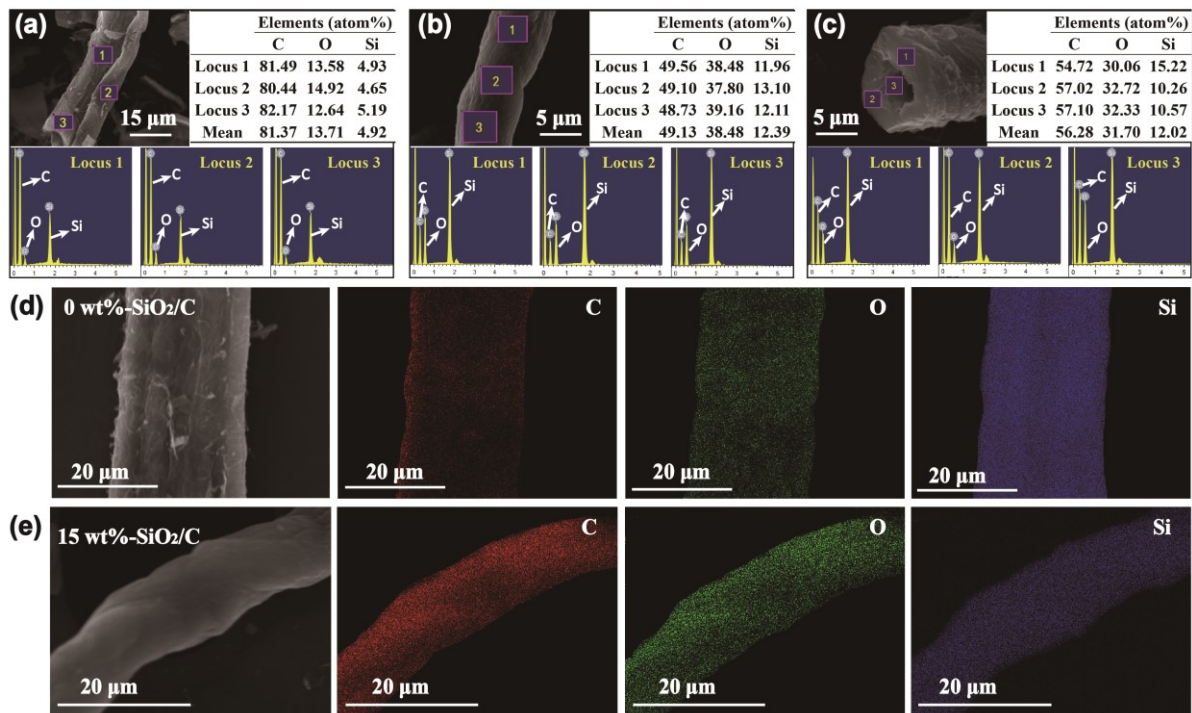
233 al., 2014; Yoo et al., 2012). The N₂ absorption-desorption isotherm and pore size distribution curves
234 of these two kinds of SiO₂/C composites, *i.e.*, 0 and 15 wt%-SiO₂/C composite, are presented in
235 Fig. 2e and f, and the porous properties of these SiO₂/C composites are summarized in Table S1.
236 15 wt%-SiO₂/C composite presented a relatively higher SSA and a total pore volume than those of
237 0 wt%-SiO₂/C composite. This could be attributed to the fact that the surface of 15 wt%-SiO₂/C
238 was rougher than that of 0 wt%-SiO₂/C (insets in Fig. 2a and c). In addition, although Table S1 and
239 the insets in Fig. 2e and f reveal that these two kinds of the composite contained micro- (< 2 nm)
240 and meso-pores (2–50 nm), and the 15 wt%-SiO₂/C composite exhibited a higher V_{meso}/V_{total} ratio
241 (92.81%). The high SSA for 15 wt%-SiO₂/C composite could facilitate the infiltration of liquid
242 electrolyte into the composite electrode and thus promoted the charge transfer process (Hou et al.,
243 2016; Shen et al., 2018), and sufficient pore volume is beneficial to accommodate the volumetric
244 expansion of silica (Kim et al., 2018a). The thin and porous cell wall and large lumen of 15 wt%-
245 SiO₂/C composite might improve the electrochemical energy storage (Li et al., 2013; Mhamane et
246 al., 2018; Yun et al., 2015), and accelerated the liquid-electrolyte penetration, ion diffusion and
247 transfer (Mhamane et al., 2018). The stable tubular structure could be beneficial to forming a stable
248 solid-electrolyte-interphase (SEI) film, preventing irreversible insertion reaction of lithium-ion into
249 the carbon structure (Cao et al., 2012; Tao et al., 2017; Yao et al., 2003).



250
 251 **Fig. 2.** SEM images of the resultant SiO₂/C composites for (a) 0 wt%-SiO₂/C and (c) 15 wt%-SiO₂/C with
 252 different magnification, and the cross section of (b) 0 wt%-SiO₂/C and (d) 15 wt%-SiO₂/C; N₂ absorption-
 253 desorption isotherm and pore size distribution curves (inner plot) for (e) 0 wt%-SiO₂/C composite and (f) 15
 254 wt%-SiO₂/C composite.

255 The hypothesis, *i.e.*, high WRV of wood pulp fibers enhancing the retention and uniform
 256 distribution of SiO₂ in the target SiO₂/C composite, was also confirmed by the EDX peaks of C, O,
 257 and Si element in the SiO₂/C composites, as shown in Fig. 3. The surface of 15 wt%-SiO₂/C
 258 composite presented more Si element (12.39%) than that of the control sample (4.92%), as shown
 259 in Fig. 3a-b. This demonstrates the fact that the SiO₂ load in SiO₂/C composites was correlated
 260 with the cellulose accessibility of wood pulp fibers. The higher the cellulose accessibility, the
 261 greater the SiO₂ load. Moreover, both the axial and cross-sectional surfaces of 15 wt%-SiO₂/C
 262 composite presented almost the same Si element distribution (Fig. 3b-c), indicating almost the same

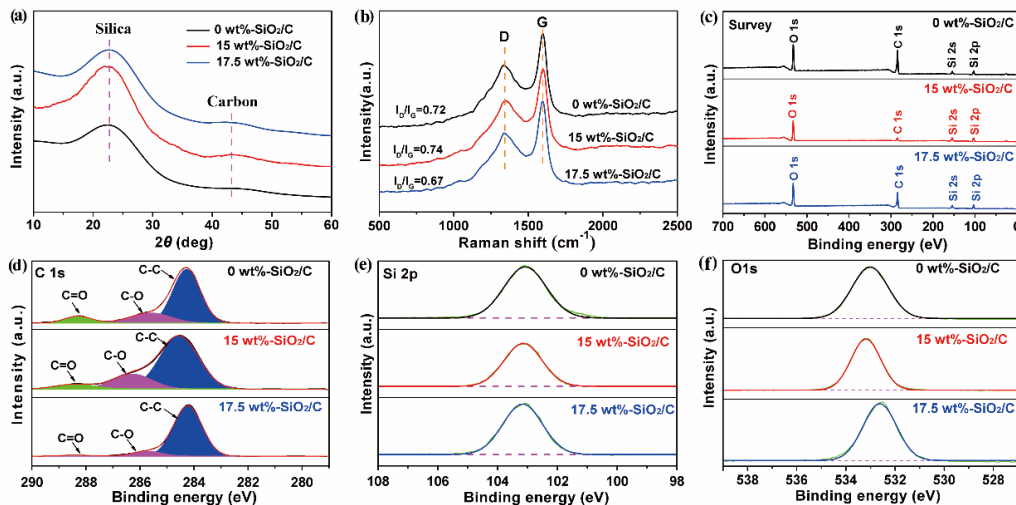
263 silicon accessibility to the axial and cross-section surfaces of the cell wall for NaOH-swollen wood
 264 pulp fibers. It is indicated that alkali pretreatment would promote the homogeneous distribution of
 265 the silicon element on the surface and inner wall of the tubular SiO₂/C composite after
 266 carbonization. When the composite is used as the anode material for LIBs, both inner and outer
 267 silica layers could participate in the lithiation reaction to accelerate liquid-electrolyte penetration
 268 and Li-ion charge/discharge. The elemental mappings (Fig. 3d-e) indicate that C, O and Si elements
 269 were all distributed uniformly throughout the SiO₂/C composite. It can be summarized that it is
 270 feasible to increase the silica content and make it uniformly distributed throughout the SiO₂/C
 271 composite prepared from NaOH-swollen wood pulp fibers by means of alkali pretreatment.



272
 273 **Fig. 3.** Axial EDX patterns for the composites of (a) 0 wt%-SiO₂/C and (b) 15 wt%-SiO₂/C; the cross-sectional EDX
 274 patterns for (c) 15 wt%-SiO₂/C composite; EDX elemental mappings of C, O, and Si in (d) 0 wt%-SiO₂/C
 275 composite and (e) 15 wt%-SiO₂/C composite.

276 Fig. 4a presents the XRD patterns of as-prepared 0, 15 and 17.5 wt%-SiO₂/C composites. All
277 the composites possessed amorphous or low crystallinity features due to no sharp diffraction peaks
278 in the XRD patterns. A broad peak located in 2θ range of 16–30° was associated with amorphous
279 SiO₂, and the weak diffraction peak around 43° was ascribed to disordered carbon (Kim et al.,
280 2018b; Li and Zhou, 2012). There was no obvious difference in the XRD pattern between 0 and
281 17.5 wt%-SiO₂/C composites except for the fact that 15 wt%-SiO₂/C composite exhibited a strong
282 diffraction peak arising from amorphous SiO₂ (Fig. 4a). In addition, there was no distinguishable
283 diffraction peak in the XRD patterns attributing to the Si, indicating that the SiO₂ wouldn't be
284 reduced to Si during the process of preparing SiO₂/C composite. It has been proved that amorphous
285 SiO₂ has higher electrochemical activity than SiO₂ with crystalline cluster (Lv et al., 2013). The
286 Raman spectra of three kinds of SiO₂/C composites (Fig. 4b) clearly show the D-band around 1360
287 cm⁻¹ and the G-band around 1590 cm⁻¹, demonstrating the existence of disordered carbon and sp²
288 graphitic carbon in the SiO₂/C composites, respectively (Mhamane et al., 2018; Zhu et al., 2017).
289 The intensity ratios of D-band to G-band (I_D/I_G) for the selected three kinds of SiO₂/C composites
290 were 0.72, 0.74 and 0.67, respectively, suggesting that the carbon in the SiO₂/C composites was
291 endowed with a relatively high degree of graphitization due to the presence of SiO₂ in the
292 composites, which could promote the growth of carbon upon heating (Agyeman et al., 2016; Shen
293 et al., 2018). By far a carbon scaffold for improving the electrical conductivity of the prepared
294 SiO₂/C composite is fabricated by using wood pulp fibers, regardless of whether wood pulp fibers
295 were pretreated with or without NaOH solution. XPS was selected to further detect the surface
296 chemical composition and chemical states of the prepared SiO₂/C composites, as shown in Fig. 4c-

297 f. The survey spectra (Fig. 4c) of each tested SiO₂/C composite contained four distinct bands at
 298 103, 153, 284, and 532 eV, corresponding to Si 2p, Si 2s, C 1s, and O 1s, respectively, which
 299 indicates that silicon, carbon and oxygen were three dominant elements in the prepared SiO₂/C
 300 composites. The high-resolution C 1s spectrum of each SiO₂/C composite could be fitted to three
 301 bands by applying the curve-fitting procedure, as shown in Fig. 4d. The C1s peaks located at 284.8,
 302 286.2 and 288.8 eV were attributed to the free carbon, C-O and C=O bonds in the SiO₂/C
 303 composites, respectively. Almost disappeared bands attributed to C-O and C=O bonds in the 17.5
 304 wt%-SiO₂/C composite indicated that alkali pretreatment of the wood pulp fibers had a great
 305 influence on the formation of oxygen functional groups in the carbonization process. Typical high-
 306 resolution Si 2p and O 1s spectra of the selected three kinds of the composites are depicted in Fig.
 307 4e and f. The bands at 103.51 eV for Si 2p and 532.5 eV for O 1s mean the silicon element in the
 308 form of silicon dioxide existed in the prepared SiO₂/C composites (Chou et al., 2018; Xiao et al.,
 309 2019). It is again confirmed that SiO₂ could not be reduced to Si in the preparation of the SiO₂/C
 310 composite.



311

312 **Fig. 4.** (a) XRD patterns, (b) Raman spectra, (c) XPS survey spectra, (d) high-resolution C 1s XPS spectra, (e)
313 high-resolution Si 2p XPS spectra, and (f) high-resolution O 1s XPS spectra of the resultant SiO₂/C composites.

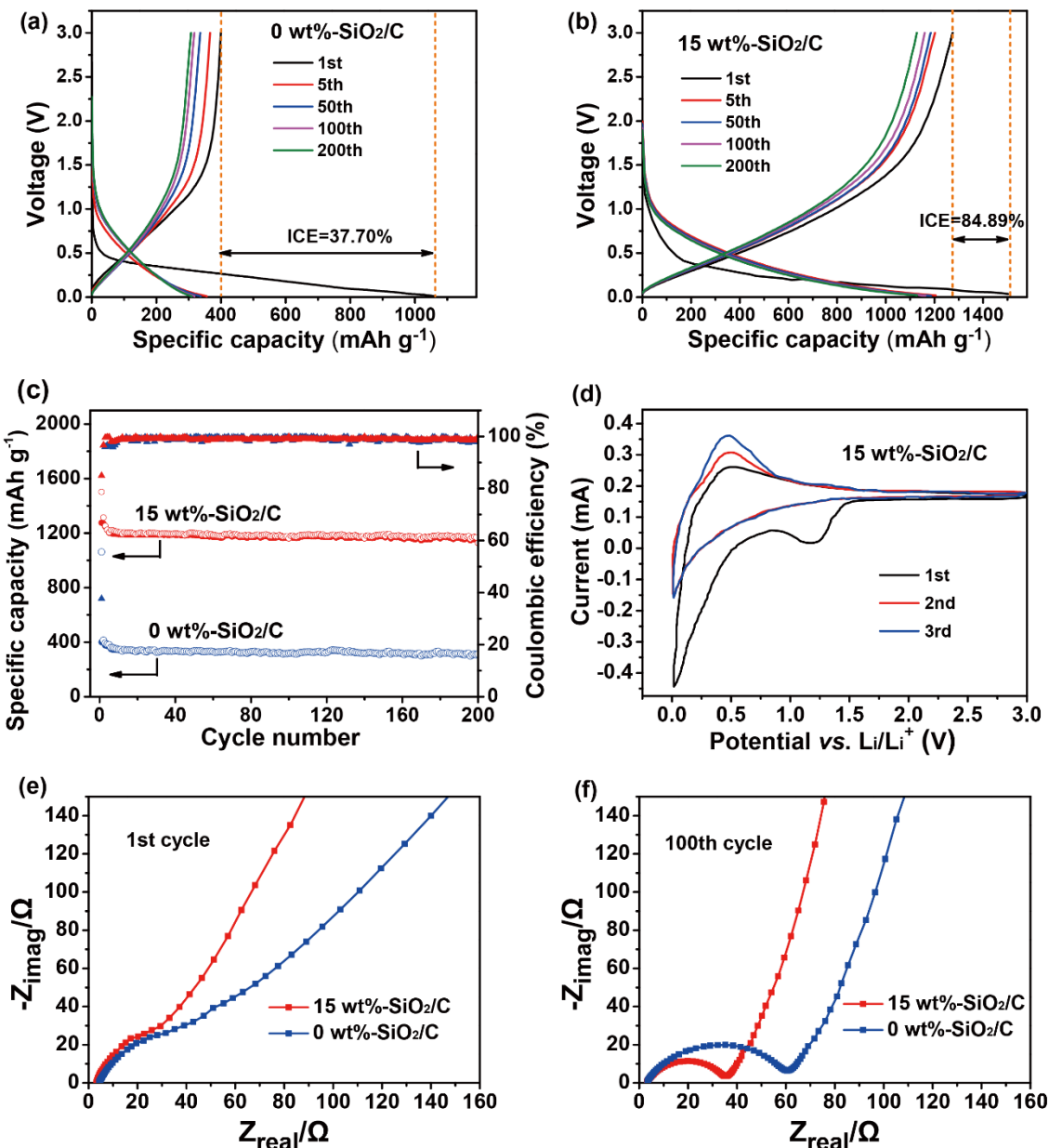
314 3.3. *Electrochemical properties of the SiO₂/C composites*

315 The cycling stability for 0 and 15 wt%-SiO₂/C composites was evaluated *via* galvanostatic
316 charging/discharging in the potential range of 0.01–3.0 V at a constant current density of 100 mA
317 g⁻¹. As shown in Fig. 5a and b, 0 and 15 wt%-SiO₂/C composites exhibited charge/discharge
318 capacity of 400/1060 and 1275/1502 mAh g⁻¹, respectively in the first cycle. The capacity loss in
319 the initial cycle can be attributed to the irreversible consumption of electrolyte, including the
320 formation of the SEI film on the surface of electrode, the reduction of SiO₂ and the formation of
321 elemental silicon and a series of silicates (Shen et al., 2018; Yang et al., 2017). The initial
322 coulombic efficiency (CE) of 15 wt%-SiO₂/C composite was 84.9% at 100 mA g⁻¹, much higher
323 than that of 0 wt%-SiO₂/C composite (37.7%), as well as other Si/C composites displayed in Table
324 1. This is attributed to the short lithium diffusion distance and a high mesoporous proportion for
325 the tubular structure of porous carbon derived from the NaOH-swollen wood pulp fibers, thus
326 reducing the consumption of lithium ions to form the SEI film and the insertion of irreversible
327 lithium into SiO₂/C composite (Kim et al., 2018b; Li and Zhou, 2012). Moreover, the
328 charge/discharge curves for 0 wt%-SiO₂/C composite remained similar in shape with a small
329 decrease in the capacity from the 5th to the 200th cycle (Fig. 5a), being similar to those for 15 wt%-
330 SiO₂/C composite (Fig. 5b). The average CE from the 5th to the 200th cycle for both 0 and 15 wt%-
331 SiO₂/C composites reached 98.8%, indicating that these two kinds of SiO₂/C composites presented

332 an excellent cycling stability (Fig. 5c). This is ascribed to the carbon derived from the wood pulp
333 fibers that could provide a reliable buffer layer and good mechanical support to the SiO₂ layer
334 (Shen et al., 2018) whether or not the wood pulp fibers were pretreated with NaOH solution. The
335 reversible specific capacity of 15 wt%-SiO₂/C composite (1130 mAh g⁻¹) was much higher than
336 that of 0 wt%-SiO₂/C composite (307 mAh g⁻¹) after 200 cycles due to a higher SiO₂ content of
337 the former. Moreover, the electrochemical performance of 15 wt%-SiO₂/C composite was also
338 comparable to those of recently reported SiO_x/C (0 ≤ x ≤ 2) anode materials with cellulose as
339 carbon resource (Table 1). The galvanostatic charge/discharge voltage profiles of 15 wt%-SiO₂/C
340 composite at 400 mA g⁻¹ is shown in Fig. S2.

341 Fig S3 and Fig. 5d show the CV curves of the first three cycles for 0 and 15 wt%-SiO₂/C
342 composites, respectively in the potential range of 0.0–3.0 V vs. Li/Li⁺ at a scanning rate of 1.0 mV
343 s⁻¹. According to previous studies (Favors et al., 2014; Jiang et al., 2013; Li et al., 2015; Schroder
344 et al., 2012), the reduction peaks at about 1.25 and 0.43 V in the first cathodic sweep can be all
345 ascribed to the decomposition of electrolyte on the composite surface, the irreversible formation of
346 the SEI film, the reduction of SiO₂, and the formation of elemental silicon and a series of silicates.
347 There is more conspicuous peak at about 0.43 V in the 0 wt%-SiO₂/C composite than that in the
348 15 wt%-SiO₂/C composite. It means that more irreversible loss of capacity during the first cycle
349 happened in the 0 wt%-SiO₂/C composite, which could be also concluded by the galvanostatic
350 charging/discharging curves (Fig. 5a and b). Notably, the reduction peak around 1.25 and 0.43 V
351 in the CV curves of all SiO₂/C composites disappeared in the subsequent curves (Fig S3 and Fig.
352 5d), suggesting that stable SEI film was formed for these two kinds of SiO₂/C composites after the

353 first cycle. When the potential got lower to 0.1 V in the cathodic scan, a sharp reduction peak could
 354 be observed, which is attributed to the lithiation of silicon to form a Li_xSi alloy (Dirican et al., 2015;
 355 Kim et al., 2015). The charging branch showed a broad oxidation peak around 0.5 V due to the
 356 extraction of ions from Li_xSi . Obviously, the oxidation peak exhibited an enhanced intensity during
 357 the anodic sweep, which can be mainly attributed to the gradual activation process of anode.



358

359 **Fig. 5.** Galvanostatic charge/discharge voltage profiles of (a) 0 wt%-SiO₂/C composite and (b) 15 wt%-SiO₂/C
 360 composite; (c) cycling performance of 0 wt%-SiO₂/C and 15 wt%-SiO₂/C composites at 100 mA g⁻¹; (d) CV
 361 curves of 15 wt%-SiO₂/C composite; and Nyquist plots for 0 and 15 wt%-SiO₂/C composites after (e) the 1st and
 362 (f) the 100th charge/discharge cycles.

363 **Table 1** Comparison of electrochemical performance for various SiO_x/C anode materials.

Cellulose form	Materials	Electrochemical performance			References
		Current density	Reversible specific capacity	Initial CE	
Cellulose solution	Core-shell structured Si@SiO ₂ /C composite	420 mA g ⁻¹	1071 mA h g ⁻¹ after 200 cycles	61.0% at 100 mA g ⁻¹	(Shen et al., 2018)
Cellulose nanofibers	Si/C composite with network structure	400 mA g ⁻¹	1594 mA h g ⁻¹ after 200 cycles	78.8% at 100 mA g ⁻¹	(Kim et al., 2017)
Cellulose nanofibers	textile-like carbon wrapped Si/C composite	2 A g ⁻¹	1097 mA h g ⁻¹ after 200 cycles	73.8% at 100 mA g ⁻¹	(Kim et al., 2018b)
Filter paper	Nanofibrous SiO ₂ /C composite	100 mA g ⁻¹	400.4 mA h g ⁻¹ after 200 cycles	45.9% at 100 mA g ⁻¹	(Jia et al., 2017b)
NaOH-swollen wood pulp fiber	Tubular SiO ₂ /C composite	100 mA g ⁻¹	1130 mA h g ⁻¹ after 200 cycles	84.9% at 100 mA g ⁻¹	This work
		400 mA g ⁻¹	1056.4 mA h g ⁻¹ after 200 cycles	76.02% at 400 mA g ⁻¹	

364 Electrochemical impedance spectroscopy (EIS) tests were employed to explore the impedance
 365 of 0 and 15 wt%-SiO₂/C composites at a frequency range from 0.1 MHz to 0.01 Hz at an amplitude
 366 ratio of 5 mV. Nyquist plots at the fresh status and the 100th cycle for 0 and 15 wt%-SiO₂/C
 367 composites were respectively obtained. As shown in Fig. 5e, the depressed semicircle could not be
 368 observed from the curves of 0 and 15 wt%-SiO₂/C composites at fresh status, which resulted from
 369 the inexistence of the SEI film in fresh cells. After 100 cycles (Fig. 5f), however, 15 wt%-SiO₂/C

370 composite exhibited a smaller semicircle at the medium-frequency zone, suggesting a smaller
371 charge transfer resistance on the electrode/electrolyte as compared to that of 0 wt%-SiO₂/C
372 composite (Kim et al., 2015). Overall, the alkaline swelling of wood pulp fibers would endow the
373 resultant SiO₂/C composites with an excellent electrochemical performance.

374 4. Conclusions

375 The carbon skeleton of the SiO₂/C composite derived from wood pulp fibers could provide a
376 reliable buffer layer and good mechanical support to the SiO₂ layer whether or not wood pulp fibers
377 were pretreated with NaOH solution, which could endow the SiO₂/C composite used as the anode
378 material for LIBs with excellent cycling stability. The resultant SiO₂/C composite remained the
379 morphology of wood pulp fibers, and the tubular SiO₂/C composite derived from NaOH-swollen
380 wood pulp fibers, *e.g.*, 15 wt%-SiO₂/C composite, had a silica load of 12.39% and a reversible
381 specific capacity of 1130 mAh g⁻¹ at 100 mA g⁻¹ after 200 cycles, behaving better performance
382 than the reference, *i.e.*, 0 wt%-SiO₂/C composite (307 mAh g⁻¹). The improvement is proposed
383 with the association of the improved accessibility of wood pulp fibers to the liquid silica precursor
384 in the process of alkaline swelling. High mesoporous proportion and the tubular morphology of
385 prepared 15 wt%-SiO₂/C composite could shorten the lithium diffusion distance and reduce the
386 consumption of lithium ions to form the SEI film obtaining a high initial coulombic efficiency.
387 This work demonstrated the feasibility of regulating the morphology of wood pulp fibers by
388 pretreatment, *e.g.*, alkali pretreatment, in order to obtain SiO₂/C composites with ideal morphology
389 and excellent electrochemical performance.

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393 **Appendix A. Supplementary data**

394 Supplementary data associated with this article can be found, in the online version.

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